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Name:		e-mail:		

#### OCR AS Level Chemistry — Paper 1 Unit 1: Atoms, Bonds and Groups (Each question = 1 mark)

- 1. Which one of the following statements about the atom is fulse?
- A neutron has the same relative mass as a proton.
- B The number of neutrons in an atom is the same as the number of pretons.
- C The relative mass of an electron is very small compared to that of a proton
- D Clouds of negative charge surround the nucleus.

В 2.4%

- 2. The number of neutrons present in the cobelt isotope. Co is
- A 33
- B 27
- C 60 D 87

A .



 The analysis of a sample of sphidium using a mass spectrometer showed that two isotope were present. S Rb and S Rb. The table below shows the percentage abundance for each isotope:

a Tenda isotope	Rb	<sup>87</sup> Rb
relative isotopic mass	85.0	87.0
% abundance	72.1	27.9

Which one of the following is the relative atomic mass of rubidium to three significant figures?

- A 85.5
- B 86.5
- C 85.6 D 86.4





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