

Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON_METALS share electrons, attempting to attain a stable octet (8 outer electrons) in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

| |
|--|
| <p>1.) H₂ hydrogen is diatomic:</p> $\text{H} + \text{H} \longrightarrow \text{H} \begin{array}{c} \cdot \\ \cdot \end{array} \text{H} \longrightarrow \text{H}-\text{H} \quad \text{Single Bond}$ |
| <p>2.) F₂ fluorine is diatomic:</p> $\text{F} + \text{F} \longrightarrow \begin{array}{c} \cdot\cdot \\ \cdot\cdot \\ \cdot\cdot \\ \cdot\cdot \end{array} \text{F} \begin{array}{c} \cdot\cdot \\ \cdot\cdot \\ \cdot\cdot \\ \cdot\cdot \end{array} \text{F} \longrightarrow$ |
| <p>3.) O₂ oxygen is diatomic:</p> $\text{O} + \text{O} \longrightarrow \quad \quad \quad \longrightarrow \text{O}=\text{O} \quad \text{Double Bond}$ |
| <p>4.) N₂ nitrogen is diatomic. Is this a triple bond?</p> $\text{N} + \text{N}$ |
| <p>5.) BF₃ you need 3 fluorine atoms here:</p> $\text{B} + \text{F}$ |
| <p>6.) Ammonia NH₃ hint: how many hydrogen atoms are needed?</p> $\text{N} + \text{H}$ |
| <p>7.) Carbon dioxide CO₂</p> $\text{O} + \text{C} + \text{O}$ |
| <p>8.) Methane CH₄ careful here 4 hydrogen atoms needed</p> $\text{C} + \text{H}$ |
| <p>9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!!</p> $\text{H} + \text{O} + \text{H}$ |
| <p>10.) SO₂ hint: one pair of electrons from sulfur must be slit up for this one to work.</p> $\text{O} + \text{S} + \text{O}$ |

Exercise On Covalent Bonding Worksheet Answers

Sebastian Brünink



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