ChemActivity 33 Is Energy Released or Consumed When a Reaction Occurs?

219

Exercises

- 1. Calculate ΔH^a for each of the following reactions:
 - a) $MgO(s) + H_2O(\ell) \longrightarrow Mg(OH)_2(s)$
 - b) 2 Zn(s) + O2(g) → 2 ZnO(s)
 - c) TiCl4(g) + 2 H2O(g) TiO2(s) + 4 HCl(g)
- Given a balanced chemical equation, describe how to determine if energy is released or consumed when that chemical reaction occurs.

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