

Chemistry – Mock Test Paper

[Time: $1\frac{1}{2}$ hrs]

[M. Marks : 80]

Answers to this paper must be written on the paper provided separately.

You will not be allowed to write during the first 15 minutes.

This time is to be spent in reading the Question Paper.

The time given at the head of this paper is the time allowed for writing the answers.

Section I is compulsory. Attempt any four questions from Section II.

Section I (40 Marks)

Attempt all questions from this section

Question 1.

(a) Choose the most appropriate answer.

- (i) Elements belonging to the same group have similar properties, because :
 - A. they have similar electronic configuration in valence shell.
 - B. their atomic numbers increase as we move down a group.
 - C. they are all metallic or non-metallic elements.
 - D. their number of electrons increase steadily.
- (ii) With the increase in atomic number in a period :
 - A. the metallic character increases.
 - B. the metallic character decreases.
 - C. the chemical reactivity decreases.
 - D. the chemical reactivity increases.
- (iii) Amongst lithium, sodium and potassium the atomic size of :
 - A. All of them is same.
 - B. Sodium has the largest atomic size.
 - C. Lithium has the largest atomic size.
 - D. Potassium has the largest atomic size.
- (iv) Amongst Be, B, Si, C and Cl :
 - A. Be and B are metalloids.
 - B. B and Si are metalloids.
 - C. Si and Cl are metalloids.
 - D. Si and C are metalloids.
- (v) On moving horizontally across a period, the number of electrons in outermost shell increase from :
 - A. 2 to 8
 - B. 1 to 8
 - C. 1 to 18
 - D. 2 to 18
- (vi) Magnesium atom loses 2 electrons in its valence shell to acquire the stable configuration of nearest noble gas is :
 - A. Neon
 - B. Argon
 - C. Helium
 - D. None of these
- (vii) A compound of two non-metals which dissolves in water to form an alkali is :
 - A. methane
 - B. carbon dioxide
 - C. Phosphorous penta oxide
 - D. Ammonia.
- (viii) One twelfth mass of carbon atom $^{12}_6\text{C}$ is called :
 - A. Atomic mass
 - B. Atomic number
 - C. Atomic mass unit
 - D. All of these.
- (ix) The electrolyte used during silver plating is :
 - A. silver nitrate solution.
 - B. fused silver chloride
 - C. sodium argento cyanide solution.
 - D. all the above
- (x) Non-metals are generally :
 - A. oxidizing agents
 - B. reducing agents
 - C. bleaching agents
 - D. neutral in nature.

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