

EXPERIMENT NO: SINGLE STAGE LEACHING

Aim: To determine the efficiency of single stage leaching experiment for leaching of NaOH and CaCO_3 .

Apparatus: Glass beakers, glass rod, measuring cylinder and stirrer.

Chemicals: NaOH, CaCO_3 , Phenolphthalein and water.

Procedure:

- Clean all apparatus with water.
- Prepare the mixture for extraction by mixing 5 gm NaOH and 3.5 gm CaCO_3 and transfer it to a beaker.
- Add 200 ml of water into the beaker and stir the solution for about 10 min.
- Allow the mixture to settle and a clear liquid layer will form at the top, this is the extract layer.
- Take 25 ml of the clear solution and titrate it against 0.5 N HCl with phenolphthalein indicator.
- Note down the observation and do the calculation.

Observation:

- | | |
|-----------------------------------|---|
| 1. Weight of feed NaOH | - |
| 2. Weight of feed CaCO_3 | - |
| 3. Weight of solvent (water) | - |
| 4. Volume of clear solution | - |
| 5. Indicator | - |
| 6. Volume of extract | - |
| 7. Titrate reading | - |
| 8. Normality of HCl solution | - |

Calculation:

i. For clear solution normality, $N_1 = \frac{N_2 V_2}{V_1}$

ii. Amount of NaOH in clear solution (extract)
$$= N_1 \times \text{volume} \times \text{gm equivalent of NaOH}$$

Mto Laboratory Testing Manual

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Table of Contents Mto Laboratory Testing Manual

1. Understanding the eBook Mto Laboratory Testing Manual
 - The Rise of Digital Reading Mto Laboratory Testing Manual
 - Advantages of eBooks Over Traditional Books
2. Identifying Mto Laboratory Testing Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mto Laboratory Testing Manual
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mto Laboratory Testing Manual
 - Personalized Recommendations
 - Mto Laboratory Testing Manual User Reviews and Ratings
 - Mto Laboratory Testing Manual and Bestseller Lists
5. Accessing Mto Laboratory Testing Manual Free and Paid eBooks

- Mto Laboratory Testing Manual Public Domain eBooks
- Mto Laboratory Testing Manual eBook Subscription Services
- Mto Laboratory Testing Manual Budget-Friendly Options
- 6. Navigating Mto Laboratory Testing Manual eBook Formats
 - ePub, PDF, MOBI, and More
 - Mto Laboratory Testing Manual Compatibility with Devices
 - Mto Laboratory Testing Manual Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mto Laboratory Testing Manual
 - Highlighting and Note-Taking Mto Laboratory Testing Manual
 - Interactive Elements Mto Laboratory Testing Manual
- 8. Staying Engaged with Mto Laboratory Testing Manual
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mto Laboratory Testing Manual
- 9. Balancing eBooks and Physical Books Mto Laboratory Testing Manual
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mto Laboratory Testing Manual
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Mto Laboratory Testing Manual
 - Setting Reading Goals Mto Laboratory Testing Manual
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mto Laboratory Testing Manual
 - Fact-Checking eBook Content of Mto Laboratory Testing Manual
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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