

Guided Reading (Chapter 18 Section 2)

1. What do chemical bonds do?

Chemical bonds control the extent to which atoms combine with each other.

2. Is energy used to break or form chemical bonds? How do we know? Energy is released when bonds form.

3. Why do some molecules undergo a chemical reaction? Some molecules are unstable and react to form more stable molecules.

The rate of a chemical reaction is determined by the activation energy.

4. What are some ways to speed up a chemical reaction? Heat, light, and catalysts.

5. Is energy used to break or form chemical bonds? How do we know? Energy is released when bonds form.

6. What is a chemical reaction?

A chemical reaction is a process in which one or more substances are converted into one or more new substances.

7. Why are the bonds between atoms in a molecule called covalent bonds?

Because they are formed by the sharing of electrons.

8. How many electrons does the nucleus of a hydrogen atom have?

One electron.

9. What are the high-energy bonds between atoms called?

Covalent bonds.

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SA Adler



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