

Physical Sciences
General Development Study Materials: Chapter 8

Name _____ Class _____

Date _____

Chapter 8: Kidneys, Acids, and Bases

Preparation of 8. Liquid Electrolytes (page 184)

1. Which physical properties of an electrolyte can differ from those of the solvent and solute(s)?

- a. _____ conductivity _____
- b. _____ freezing point _____
- c. _____ boiling point _____

2. Compare the concentrations of solid sodium chloride and potassium

_____ potassium chloride and sodium chloride when equal volumes of each solution are being compared.

3. Consider the data below that identify what happened to water in the tests.

- a. The water molecules became more organized.
- b. The water molecules became more disorganized.
- c. The water molecules froze.
- d. The water molecules as large molecules in a non-polar solution.

Prep of Solutions (page 184)

4. Consider sodium hydroxide as added to

water _____ potassium _____ potassium, and is released from.

5. Consider potassium when added to

water _____ potassium _____ potassium, and is released from.

6. Explain following comments in or (page 17) regarding the differences among solid and liquid electrolytes among solid and liquid electrolytes.

7. Compare the of the solution.

Guided Reading And Study Workbook Chapter 2acids And Bases

Clifford Lane



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