

CHEMICAL BONDS

IONIC, COVALENT, & METALLIC

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READING COMPREHENSION & QUESTIONS

SCIENCE REAL

Types of Chemical Bonds Reading Comprehension

What is the difference between a molecule and a compound?

What of the substances shown in this diagram are ionic?

What of the substances shown in this diagram are covalent?

Chemical Bonds and Electronegativity

Electronegativity is an atom's ability to attract electrons. Elements with low electronegativity have outer shells that are almost empty and tend to gain or lose electrons easily. Elements with high electronegativity have outer shells that are almost full and tend to hold on to their electrons. Ionic bonds form when the electronegativity of two atoms differs significantly, and covalent bonds form when electronegativity of the atoms are more similar. Typically, the greater the difference in electronegativity between two elements, the more ionic the bond. The greater the similarity in electronegativity, the more covalent the bond.

Electronegativity increases from bottom left to top right of the periodic table.

Which atom has a higher electronegativity?

Which is a cation and which is an anion?

Which type of bond is formed between the two atoms?

GUIDED READING ACTIVITY

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M Mark



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Table of Contents Guided Pracice Page Covalent Bonds Page 72 Answers

1. Understanding the eBook Guided Pracice Page Covalent Bonds Page 72 Answers
 - The Rise of Digital Reading Guided Pracice Page Covalent Bonds Page 72 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Guided Pracice Page Covalent Bonds Page 72 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Guided Pracice Page Covalent Bonds Page 72 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Guided Pracice Page Covalent Bonds Page 72 Answers
 - Personalized Recommendations
 - Guided Pracice Page Covalent Bonds Page 72 Answers User Reviews and Ratings
 - Guided Pracice Page Covalent Bonds Page 72 Answers and Bestseller Lists

5. Accessing Guided Pracice Page Covalent Bonds Page 72 Answers Free and Paid eBooks
 - Guided Pracice Page Covalent Bonds Page 72 Answers Public Domain eBooks
 - Guided Pracice Page Covalent Bonds Page 72 Answers eBook Subscription Services
 - Guided Pracice Page Covalent Bonds Page 72 Answers Budget-Friendly Options
6. Navigating Guided Pracice Page Covalent Bonds Page 72 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Guided Pracice Page Covalent Bonds Page 72 Answers Compatibility with Devices
 - Guided Pracice Page Covalent Bonds Page 72 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Guided Pracice Page Covalent Bonds Page 72 Answers
 - Highlighting and Note-Taking Guided Pracice Page Covalent Bonds Page 72 Answers
 - Interactive Elements Guided Pracice Page Covalent Bonds Page 72 Answers
8. Staying Engaged with Guided Pracice Page Covalent Bonds Page 72 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Guided Pracice Page Covalent Bonds Page 72 Answers
9. Balancing eBooks and Physical Books Guided Pracice Page Covalent Bonds Page 72 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Guided Pracice Page Covalent Bonds Page 72 Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Guided Pracice Page Covalent Bonds Page 72 Answers
 - Setting Reading Goals Guided Pracice Page Covalent Bonds Page 72 Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Guided Pracice Page Covalent Bonds Page 72 Answers
 - Fact-Checking eBook Content of Guided Pracice Page Covalent Bonds Page 72 Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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