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OCR AS Level Chemistry — Paper 1 Unit 1: Atoms, Bonds and Groups (Each question = 1 mark)

- 1. Which one of the following statements about the atom is fulse?
- A neutron has the same relative mass as a proton.
- B The number of neutrons in an atom is the same as the number of pretons.
- C The relative mass of an electron is very small compared to that of a proton
- D Clouds of negative charge surround the nucleus.

В 2.4%

- 2. The number of neutrons present in the cobelt isotope. Co is
- A 33
- B 27
- C 60 D 87

A .



 The analysis of a sample of sphidium using a mass spectrometer showed that two isotope were present. S Rb and S Rb. The table below shows the percentage abundance for each isotope:

, Teach isotope	Rb	⁸⁷ Rb
relative isotopic mass	85.0	87.0
% abundance	72.1	27.9

Which one of the following is the relative atomic mass of rubidium to three significant figures?

- A 85.5
- B 86.5
- C 85.6 D 86.4





F321 2014 Paper

Michal Rosen-Zvi

F321 2014 Paper:

F321 2014 Paper Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has be apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "F321 2014 Paper," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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