

Name: samantha poe Date: 11.18.22

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? *Frank can share his red marker with Flora and Flora can share her blue marker with Frank.*

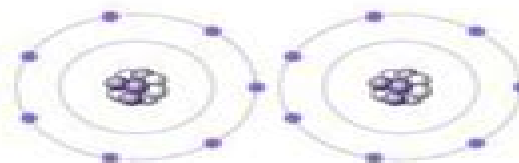
2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

Olivia can share her purple and green markers with Otto and Otto can share his black and brown markers with Olivia.

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo™, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? *Each fluorine atom has seven valence electrons.*
2. Click **Pause** (). Drag an electron from the left atom to the right atom. Click **Play**. What happens? *The electrons that was dragged now orbits both atoms.*

Exploration Guide Covalent Bonds

Julia Schneider



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Fundamentals Of Quantum Materials: A Practical Guide To Synthesis And Exploration Johnpierre Paglione, Nicholas P Butch, Efrain E Rodriguez, 2021-01-04 Despite a long tradition of sophisticated creative materials synthesis among quantum materials researchers a sense of broader community has been lacking In initiating the Fundamentals of Quantum Materials Winter School held annually at the University of Maryland we wanted to bring together the next generation of growers to learn techniques and pointers directly from senior scientists and it turns out that we were not alone The enthusiasm from both students and teachers has been both gratifying and invigorating Four schools later we can confidently say that physicists chemists and materials scientists experimentalists and theorists alike all want to know how to make a good sample With this in mind we asked our lecturers to record their most important ideas and share their expertise with a broader audience This resource is a compilation of fundamental and practical guides on the modern methods of materials synthesis utilized by these experts We hope that you enjoy reading their essential guidance and state of the art techniques as you explore the Fundamentals of Quantum Materials

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