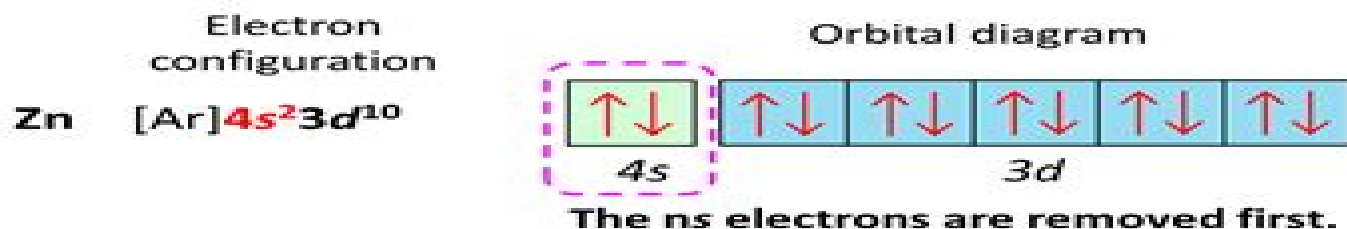
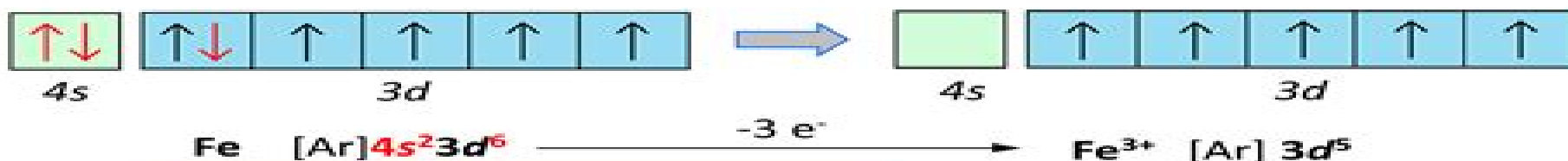


Electron Configuration of Transition Metal Cations



The electron(s) are removed **first from the ns** (highest principal quantum number) orbital and **then from the (n - 1)d** orbitals



In contrast to main-group ions, transition metal ions **do not** usually attain a noble gas configuration.



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