

Evidence for Chemical Change

One way of knowing that a chemical change has occurred is to observe that the properties of the products are different from those of the reactants. A new product can then become a reactant in another chemical reaction. In this experiment, you will observe a sequence of changes that occur when a solution that begins as copper(II) nitrate is subjected to a series of different reactions. All of the reactions will take place in the same test tube. At each step, you will look for evidence that a new substance has formed as a result of a chemical change. You will also observe energy changes and relate them to chemical reactions.

OBJECTIVES

Observe evidence that a chemical change has taken place.

Infer from observations that a new substance has been formed.

Identify and record observations that show energy is involved in chemical change.

Observe the color and solubility of some substances.

Describe reactions by writing word equations.

MATERIALS

- aluminum wire, 12 cm
- beaker, 100 mL
- Bunsen burner and related equipment
- copper(II) nitrate, 0.75 M
- glass stirring rod
- gloves
- HCl, 3.0 M
- iron ring
- lab apron
- lab marker
- NaOH, 3.0 M
- ring stand
- ruler
- safety goggles
- test tube, 13 mm × 100 mm
- test-tube rack
- wire gauze, ceramic-centered



Always wear safety goggles and a lab apron to protect your eyes and clothing. If you get a chemical in your eyes, immediately flush the chemical out at the eyewash station while calling to your teacher. Know the location of the emergency lab shower and the eyewash station and the procedures for using them.



Do not touch any chemicals. If you get a chemical on your skin or clothing, wash the chemical off at the sink while calling to your teacher. Make sure you carefully read the labels and follow the precautions on all containers of chemicals that you use. If there are no precautions stated on the label, ask your teacher what precautions to follow. Do not taste any chemicals or items used in the laboratory. Never return leftover chemicals to their original containers; take only small amounts to avoid wasting supplies.

Experiment A5 Evidence For Chemical Change

J Dewey



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Table of Contents Experiment A5 Evidence For Chemical Change

1. Understanding the eBook Experiment A5 Evidence For Chemical Change
 - The Rise of Digital Reading Experiment A5 Evidence For Chemical Change
 - Advantages of eBooks Over Traditional Books
2. Identifying Experiment A5 Evidence For Chemical Change
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Experiment A5 Evidence For Chemical Change
 - User-Friendly Interface

4. Exploring eBook Recommendations from Experiment A5 Evidence For Chemical Change
 - Personalized Recommendations
 - Experiment A5 Evidence For Chemical Change User Reviews and Ratings
 - Experiment A5 Evidence For Chemical Change and Bestseller Lists
5. Accessing Experiment A5 Evidence For Chemical Change Free and Paid eBooks
 - Experiment A5 Evidence For Chemical Change Public Domain eBooks
 - Experiment A5 Evidence For Chemical Change eBook Subscription Services
 - Experiment A5 Evidence For Chemical Change Budget-Friendly Options
6. Navigating Experiment A5 Evidence For Chemical Change eBook Formats
 - ePub, PDF, MOBI, and More
 - Experiment A5 Evidence For Chemical Change Compatibility with Devices
 - Experiment A5 Evidence For Chemical Change Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Experiment A5 Evidence For Chemical Change
 - Highlighting and Note-Taking Experiment A5 Evidence For Chemical Change
 - Interactive Elements Experiment A5 Evidence For Chemical Change
8. Staying Engaged with Experiment A5 Evidence For Chemical Change
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Experiment A5 Evidence For Chemical Change
9. Balancing eBooks and Physical Books Experiment A5 Evidence For Chemical Change
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Experiment A5 Evidence For Chemical Change
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Experiment A5 Evidence For Chemical Change
 - Setting Reading Goals Experiment A5 Evidence For Chemical Change
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Experiment A5 Evidence For Chemical Change
 - Fact-Checking eBook Content of Experiment A5 Evidence For Chemical Change
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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