

Experiment 22 Report Sheet

Molar Solubility, Common-Ion Effect

Date _____ Lab Sec. _____ Name _____ Desk No. _____

A. Molar Solubility and Solubility Product of Calcium Hydroxide

	Trial 1	Trial 2	Trial 3
1. Volume of saturated Ca(OH)_2 solution (mL)	25.0	25.0	25.0
2. Concentration of standardized HCl solution (mol/L)		0.05 M	
3. Buret reading, initial (mL)	0 mL	24.96	24.96
4. Buret reading, final (mL)	24.9	24.9	24.9
5. Volume of HCl added (mL)			
6. Moles of HCl added (mol)			
7. Moles of OH^- in saturated solution (mol)			
8. $[\text{OH}^-]$, equilibrium (mol/L)			
9. $[\text{Ca}^{2+}]$, equilibrium (mol/L)			
10. Molar solubility of Ca(OH)_2 (mol/L)			
11. Average molar solubility of Ca(OH)_2 (mol/L)			
12. K_{sp} of Ca(OH)_2			
13. Average K_{sp}			Data Analysis, B
14. Standard deviation of K_{sp}			Data Analysis, C
15. Relative standard deviation of K_{sp} (%RSD)			Data Analysis, D

*Calculations for Trial 1.

Experiment 22 Molar Solubility Lab Report

U.S. Atomic Energy Commission



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