

Experimental Design

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Chem- 119- 520

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Introduction

The background of this experiment is from Dr. Erguder and his experiment to see if different cooking techniques had effects on antioxidants found in fruit and vegetables¹. The scientific method is the way scientists collect data as it pertains to the hypothesis that they form. The scientific method was used in this experiment to collect data for the hypothesis "Different masses of vegetables produce different volumes of bubbles." Experimental design is important because it helps keep information concise and straining to the point, which makes it easier for other scientists to read and share.

Summary

Be sure to wear proper PPE and goggles while doing the experiment and dispose of waste accordingly. First, add .5g of the cabbage sample into a 25mL graduated cylinder. Next, add 20 drops of soap into the same graduated cylinder containing the cabbage. Then measure out 10mL of 3% hydrogen peroxide and pour it carefully into the graduated cylinder containing the cabbage and soap. Record initial volume. Set a timer for 3 minutes exactly. Record the final volume after the timer has gone off. Repeat these steps 2 more times and record the data. To record the volume of bubbles, subtract the initial volume from the final volume. Finally, follow these steps for the spinach, orange peel, and carrot samples.

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Experiment Design In A Report

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