

EXPERIMENT 19 Calcium Compounds and Le Chatelier's Principle

This chemical reaction had not yet reached equilibrium with respect to the calcium compounds and hydroxide ions. The reaction is summarized below:



Predictions:

Predict the new color when further added Calcium Compound reacts with hydroxide ions. Complete this table below and submit your observations about color:

https://www.youtube.com/watch?v=62556C3_7s

Data and Observation:

B. EFFECT OF CONCENTRATION		
Substances added	Observations	Shift in equilibrium (to the right, to the left, no shift)
+ Concentrated HCl	When this concentrated acid was added, the pink solution became blue almost instantly. The presence of acid ions effectively neutralized the base ions, decreasing the amount of base ions in the solution. The pink color of the solution faded, and the color faded to blue.	Shift to right
+ CaCl_2	When this white powder was added, the pink color of the solution faded, and the color faded to blue. The solution became blue almost instantly. The presence of calcium ions effectively increased the amount of calcium ions in the solution, shifting the equilibrium to the right.	Shift to right
+ H_2O	When this colorless liquid was added, the pink color of the solution faded, and the color faded to blue. The solution became blue almost instantly. The presence of water ions effectively increased the amount of water ions in the solution, shifting the equilibrium to the right.	Shift to left
	When this white powder was added, the pink color of the solution faded, and the color faded to blue. The solution became blue almost instantly. The presence of calcium ions effectively increased the amount of calcium ions in the solution, shifting the equilibrium to the right.	Shift to right

Experiment 9 Chemical Equilibrium

AW Chickering



Experiment 9 Chemical Equilibrium:

Guided Inquiry Experiments for General Chemistry Nancy K. Kerner, Ram S. Lamba, 2007-10-19 The use of the laboratory is a valuable tool in developing a deeper understanding of key chemical concepts from the experimental process This lab manual encourages scientific thinking enabling readers to conduct investigations in chemistry It shows how to think about the processes they are investigating rather than simply performing a laboratory experiment to the specifications set by the manual Each experiment begins with a problem scenario and ends with questions requiring feedback on the problem

CliffsNotes AP Chemistry Bobrow Test Preparation Services, 2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter's end Two full length practice tests with detailed answer explanations are included in the book

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sixteen discrete laboratory exercises to prepare AP Chemistry students for the required laboratory experiments section on the exam

Experiments in College Chemistry Lyman Churchill Newell, 1925 **Chemical Engineering Catalog**, 1925 *Bulletin of the Chemical Society of Japan* Nihon Kagakkai, 1928 **Studies from the Physical and Chemical Laboratories of the Owens College ...** Owens College, 1893 **Studies from the Physical and Chemical Laboratories of the Owens College ...**, 1893 **Illustrated Guide to Home Chemistry Experiments** Robert Bruce Thompson, 2012-02-17

For students DIY hobbyists and science buffs who can no longer get real chemistry sets this one of a kind guide explains how to set up and use a home chemistry lab with step by step instructions for conducting experiments in basic chemistry not just to make pretty colors and stinky smells but to learn how to do real lab work Purify alcohol by distillation Produce hydrogen and oxygen gas by electrolysis Smelt metallic copper from copper ore you make yourself Analyze the makeup of seawater bone and other common substances Synthesize oil of wintergreen from aspirin and rayon fiber from paper Perform forensics tests for fingerprints blood drugs and poisons and much more From the 1930s through the 1970s chemistry sets were among the most popular Christmas gifts selling in the millions But two decades ago real chemistry sets began to disappear as manufacturers and retailers became concerned about liability em The Illustrated Guide to Home Chemistry Experiments steps up to the plate with lessons on how to equip your home chemistry lab master laboratory skills and work safely in your lab The bulk of this book consists of 17 hands on chapters that include multiple laboratory sessions on the following topics Separating Mixtures Solubility and Solutions Colligative Properties of Solutions Introduction to Chemical Reactions Stoichiometry Reduction Oxidation Redox Reactions Acid Base Chemistry Chemical Kinetics Chemical Equilibrium and Le Chatelier's Principle Gas Chemistry Thermochemistry and Calorimetry Electrochemistry Photochemistry Colloids and Suspensions Qualitative Analysis Quantitative Analysis Synthesis of Useful Compounds Forensic Chemistry With plenty of full color illustrations and photos Illustrated Guide to Home Chemistry Experiments offers introductory level sessions suitable for a middle school or first year high school chemistry laboratory course and more advanced sessions suitable for students who intend to take the College Board Advanced Placement AP Chemistry exam A student who completes all of the laboratories in this book will have done the equivalent of two full years of high school chemistry lab work or a first year college general chemistry laboratory course This hands on introduction to real chemistry using real equipment real chemicals and real quantitative experiments is ideal for the many thousands of young people and adults who want to experience the magic of chemistry

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Ion Mobility-Mass Spectrometry Alison E Ashcroft, Frank Sobott, 2021-11-11 Over the last decade the use of ion mobility separation in combination with mass spectrometry analysis has developed significantly This technique adds a unique extra dimension enabling the in depth analysis of a wide range of complex samples in the areas of the chemical and biological sciences Providing a comprehensive guide to the technique each chapter is written by an internationally recognised expert and with numerous different commercial platforms to choose from this book will help the end users understand the practicalities of using different instruments for different ion mobility purposes The first section provides a detailed account of the fundamentals behind the technique and the current range of available instrumentation The second section focusses on the wide range of applications that have benefitted from ion mobility mass spectrometry and includes topics taken from current research in the pharmaceutical metabolomics glycomics and structural molecular biology fields The book is primarily aimed at researchers appealing to practising chemists and biochemists as well as those in the pharmaceutical and medical fields Journal of the American Chemical Society American Chemical Society, 1921 Proceedings of the Society are included in v 1 59 1879 1937

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