Name : \_\_\_\_\_\_\_\_ Date: \_\_\_\_\_\_

# **Chemical Reactions**

A. Balance the following chemical reactions:

3) NaCl + 
$$F_2 \longrightarrow NaF + Cl_2$$

5) 
$$CH_4 + O_2 \longrightarrow CO_2 + H_2O$$

6) 
$$H_2SO_4 + B(OH)_3 \longrightarrow B_2(SO_4)_3 + H_2O$$

7) 
$$C_5H_9O$$
 +  $O_2$   $\longrightarrow$   $CO_2$  +  $H_2O$ 

8) 
$$\text{Li}_3\text{N}$$
 +  $\text{NH}_4\text{NO}_3$   $\longrightarrow$   $\text{LiNO}_3$  +  $(\text{NH}_4)_3\text{N}$ 

10) Pb + 
$$H_3PO_4$$
  $\longrightarrow$   $Pb_3(PO_4)_2$  +  $H_2$ 

B. Identify the type of reaction as synthesis, decomposition, single-replacement, double-replacement, and combustion:

# **Experiment 14 Types Of Chemical Reactions Key**

**N Noddings** 

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