

SOLARLINK FR360 shortwave

Self-Powered Digital AM/FM Shortwave
Radio with Solar Power, Flashlight,
and USB Cell Phone Charger.



Features:

- Digital tuner with AM (520-1710kHz) & FM (87-108MHz)
- Shortwave
- Built-in 4 white LED light source
- 1 Flashing red LED
- Can be powered from four different sources:
 - 1) Dynamotrend crank which charges the Ni-MH battery
 - 2) From 3 AAA batteries (not included)
 - 3) AC Power
 - 4) Solar power which charges Ni-MH battery
- Connectors, with rubber gaskets/plugs, to seal out moisture: 3.5 mm headphone output, DC-in, USB phone-charger and Aux-in/4 input
- Telescopic antenna provided for maximum range and reception
- USB cell phone charger (USB cable not included)
- 3.5 mm headphone output
- Dimensions: 6.25" x 6.5" x 2.75" and 1158.75 x 165 x 69.85 mm (W x H x D)
- Weight: 1 lb. and 450 g
- Accessories: owner's manual, warranty card

* Specifications subject to change



Receives
AM/FM Bands



Digital
Tuning



Crank
Power



Solar
Powered



USB Cell Phone
Charger



LED
Flashlight



Headphone
Output



Aux-In/4
Input



USB
Compatible



DC-In
(+5V)



Telescopic
Antenna



Headphone
Out

eTón
CORPORATION

Eton Fr360 Radio Manual

Puja Mehta



Eton Fr360 Radio Manual:

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The Radio Manual ... Third Edition George Edward Sterling,1938 **Specialized Home Radio Manual** John Francis
Rider,1954 *Auto Radio Manual* inc. Sams (Howard W.) and Company,1962 Specialized Auto Radio Manual John
Francis Rider,1935 Auto Radio Manual ,1958 Radio Manual ,1964 *Official Short Wave Radio Manual* ,1934
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Whispering the Techniques of Language: An Psychological Quest through **Eton Fr360 Radio Manual**

In a digitally-driven earth where monitors reign great and quick conversation drowns out the subtleties of language, the profound techniques and emotional subtleties concealed within phrases usually go unheard. Yet, set within the pages of **Eton Fr360 Radio Manual** a interesting fictional value sporting with raw feelings, lies an extraordinary quest waiting to be undertaken. Written by a skilled wordsmith, this wonderful opus attracts viewers on an introspective trip, gently unraveling the veiled truths and profound influence resonating within the very fabric of every word. Within the emotional depths of the moving review, we shall embark upon a heartfelt exploration of the book is key themes, dissect its charming writing type, and succumb to the powerful resonance it evokes serious within the recesses of readers hearts.

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for the nitration 4.01 ml of nitric acid 4.00 ml of sulfuric acid and 4.56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4.56 ml of bromobenzene was used the amount of product predicted to be recovered was 8.7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction e.g. particularly nitration of methyl benzoate to give methyl m-nitrobenzoate thus determine the weight, melting point and percentage yield of the pure methyl m-nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene. Bromine substituted on an aromatic ring is an ortho/para directing group. It will direct an incoming electrophile to the ortho and para positions. It is also deactivating, meaning that its reactions will be slower than the identical reactions of benzene.

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Section 20711 Lab Experiment: Nitration of Bromobenzene. Introduction: The chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule. Nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid. This mixture is useful to obtain the active nitronium ion.

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture HNO_3 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by the nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic

substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group