

13.2**THE NATURE OF LIQUIDS****Section Review****Objectives**

- Identify factors that determine physical properties of a liquid
- Define *evaporation* in terms of kinetic energy
- Describe the equilibrium between a liquid and its vapor
- Identify the conditions under which boiling occurs

Vocabulary

- vaporization
- evaporation
- vapor pressure
- boiling point
- normal boiling point

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

Liquids are much 1 than gases. Liquids and solids are known as 2 states of matter. The conversion of a liquid to a gas or vapor is called 3. When a liquid that is not 4 changes to a gas, the process is called evaporation. A liquid evaporates faster when heated; however, evaporation itself is a 5 process. When a partially filled container of liquid is sealed, some of the particles at the 6 of the liquid vaporize. These particles collide with the walls of the container, producing a force called 7. The vapor pressure of a liquid can be determined by a device called a 8. A liquid boils when its 9 equals the external pressure. The normal boiling point of a liquid is the temperature at which the vapor pressure is equal to 10.

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____
8. _____
9. _____
10. _____

Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- _____ 11. Particles in a liquid don't have enough kinetic energy to overcome the attractive forces between them and vaporize.

Nature Of Liquids Section Review Answers

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Water Science for Food Health Zeki Berk, 2001-08-17 Was held in Zichron Yaakov Israel on 16-21 September 2000. This book is the complete collection of the invited lectures presented at the Symposium. Research interest in the fundamental properties of water and water-mediated phenomena has generated a wealth of new knowledge and novel technological applications in widely diverse areas: food science, agriculture, microbial and plant physiology and metabolism, human health, environment. The title of the Symposium, Water Science for Food Health, Agriculture and Environment, was defined so as to reflect this diversity. True to the tradition set by previous ISOPOW meetings, ISOPOW 2000 was planned according to the Gordon Conference concept. About 110 participants, not including accompanying guests and students from 21 different countries, attended the Symposium. The scientific programme consisted of 22 invited lectures, 60 poster presentations and two sessions of expert panel discussions with audience participation. Two additional sessions were devoted to the oral presentation and discussion of selected posters. Sufficient time for discussion was allocated at each session and between sessions and this was utilized well by the participants. The scientific level of the program met with extreme participant satisfaction. One of the most significant events of the Symposium was the panel discussion session entitled 'Critical Issues Related to Water Activity'. The original views and ideas expressed at this session and the heated discussion around controversial issues and open questions should lead to the following conclusion: Water Science will continue to generate new material of utmost importance and high scientific quality for many more ISOPOW meetings to come.

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