



A-LEVEL CHEMISTRY

MARK SCHEME

Practice Paper 1

Maximum marks: 105

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1	(a)	0)	[H-][OH-] <i>OR</i> [H ₂ O-][OH-] <i>Ignore (aq) Must have [] not ()</i>	1
		(0)	√3.46 × 10 ⁻¹⁴ (= 1.86 × 10 ⁻⁷) If no square root, CE= 0	2
			pH = 6.73 Must be 2dp	
		(010)	[H-] = 10 ^{-11,36} (= 4,365 × 10 ⁻¹² OR 4,37 × 10 ⁻¹³) Mark for working	
			$Kw = [4.365 \times 10^{-12} \ OR \ 4.37 \times 10^{-12} \times 0.047] = 2.05 \times 10^{-13}$ Allow $2.05 \times 10^{-13} = 2.1 \times 10^{-13}$ Mark for answer Ignore units	1
	(b)	-00	HCOOH - H-	11
		355	Must have but ignore brackets.	
			OR HCOOH + H ₀ O ————————————————————————————————————	
		(66)	K [H-][HCOO-] OR [H-O-][HCOO-]	1
			Must have all brackets but allow () Must be HCOOH etc. Allow ecf in formulae from (b)(i)	
		(111)	M1	1
			$K_s = \frac{[H^-]^2}{[HCOOH]} ([H^+]^2 = 1.78 \times 10^{-4} \times 0.056 = 9.97 \times 10^{-6})$	
			Allow HA or HX etc.	
			Allow $[H^*] = \sqrt{(Ka \times [HA])}$ for M1	
			M2 [H ⁺] = 3.16 × 10 ⁻³	
			Mark for answer	392
			M3 pH = 2.50 allow more than 2 dp but not fewer	1
			Allow correct pH from their wrong [H-] here only If square root shown but not taken, pH = 5.00 can score max 2 for M1 and M3	

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Didier Musso



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