

Exercises on Chemical Equilibria

1. **Equilibrium**
 A mixture containing 0.100 mol of SO_2 and 0.100 mol of O_2 is placed in a 1.00 L container. The reaction is as follows:
- $$2\text{SO}_2(g) + \text{O}_2(g) \rightleftharpoons 2\text{SO}_3(g)$$
- At equilibrium, the concentration of SO_2 is 0.040 M. Calculate the equilibrium constant K_c for this reaction.
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Lab 46 Observing Chemical Equilibrium

M Carnoy



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Lab 46 Observing Chemical Equilibrium Book Review: Unveiling the Power of Words

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