

### Producing Products in a Virtual Environment

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- [illegible]

- |   |  |
|---|--|
| 1. $\text{BaCO}_3(s) \xrightarrow{\Delta} \text{BaO}(s) + \text{CO}_2(g)$   | Reaction Type(s): <u>Decomposition</u>       |
| 2. $\text{Pb}(\text{NO}_3)_2(aq) + \text{Na}_2\text{CO}_3(aq) \rightarrow \text{PbCO}_3(s) + 2\text{NaNO}_3(aq)$      | Reaction Type(s): <u>Double Displacement</u> |
| 3. $\text{Zn}(s) + 2\text{HCl}(aq) \rightarrow \text{ZnCl}_2(aq) + \text{H}_2(g)$                                     | Reaction Type(s): <u>Single Displacement</u> |
| 4. $\text{Al}(\text{OH})_3(s) + 3\text{HNO}_3(aq) \rightarrow \text{Al}(\text{NO}_3)_3(aq) + 3\text{H}_2\text{O}(l)$  | Reaction Type(s): <u>Acid-Base</u>           |
| 5. $\text{Fe}(s) + \text{CuSO}_4(aq) \rightarrow \text{FeSO}_4(aq) + \text{Cu}(s)$                                    | Reaction Type(s): <u>Single Displacement</u> |
| 6. $\text{NaOH}(aq) + \text{HCl}(aq) \rightarrow \text{NaCl}(aq) + \text{H}_2\text{O}(l)$                             | Reaction Type(s): <u>Acid-Base</u>           |
| 7. $\text{BaCl}_2(aq) + \text{Na}_2\text{SO}_4(aq) \rightarrow \text{BaSO}_4(s) + 2\text{NaCl}(aq)$                   | Reaction Type(s): <u>Double Displacement</u> |
| 8. $\text{Fe}(s) + \text{CuSO}_4(aq) \rightarrow \text{FeSO}_4(aq) + \text{Cu}(s)$                                    | Reaction Type(s): <u>Single Displacement</u> |
| 9. $\text{Al}(\text{OH})_3(s) + 3\text{HNO}_3(aq) \rightarrow \text{Al}(\text{NO}_3)_3(aq) + 3\text{H}_2\text{O}(l)$  | Reaction Type(s): <u>Acid-Base</u>           |
| 10. $\text{BaCO}_3(s) \xrightarrow{\Delta} \text{BaO}(s) + \text{CO}_2(g)$  | Reaction Type(s): <u>Decomposition</u>       |
| 11. $\text{Zn}(s) + 2\text{HCl}(aq) \rightarrow \text{ZnCl}_2(aq) + \text{H}_2(g)$                                    | Reaction Type(s): <u>Single Displacement</u> |
| 12. $\text{Pb}(\text{NO}_3)_2(aq) + \text{Na}_2\text{CO}_3(aq) \rightarrow \text{PbCO}_3(s) + 2\text{NaNO}_3(aq)$     | Reaction Type(s): <u>Double Displacement</u> |
| 13. $\text{Fe}(s) + \text{CuSO}_4(aq) \rightarrow \text{FeSO}_4(aq) + \text{Cu}(s)$                                   | Reaction Type(s): <u>Single Displacement</u> |
| 14. $\text{Al}(\text{OH})_3(s) + 3\text{HNO}_3(aq) \rightarrow \text{Al}(\text{NO}_3)_3(aq) + 3\text{H}_2\text{O}(l)$ | Reaction Type(s): <u>Acid-Base</u>           |
| 15. $\text{BaCO}_3(s) \xrightarrow{\Delta} \text{BaO}(s) + \text{CO}_2(g)$  | Reaction Type(s): <u>Decomposition</u>       |
| 16. $\text{Zn}(s) + 2\text{HCl}(aq) \rightarrow \text{ZnCl}_2(aq) + \text{H}_2(g)$                                    | Reaction Type(s): <u>Single Displacement</u> |
| 17. $\text{Pb}(\text{NO}_3)_2(aq) + \text{Na}_2\text{CO}_3(aq) \rightarrow \text{PbCO}_3(s) + 2\text{NaNO}_3(aq)$     | Reaction Type(s): <u>Double Displacement</u> |
| 18. $\text{Fe}(s) + \text{CuSO}_4(aq) \rightarrow \text{FeSO}_4(aq) + \text{Cu}(s)$                                   | Reaction Type(s): <u>Single Displacement</u> |
| 19. $\text{Al}(\text{OH})_3(s) + 3\text{HNO}_3(aq) \rightarrow \text{Al}(\text{NO}_3)_3(aq) + 3\text{H}_2\text{O}(l)$ | Reaction Type(s): <u>Acid-Base</u>           |
| 20. $\text{BaCO}_3(s) \xrightarrow{\Delta} \text{BaO}(s) + \text{CO}_2(g)$  | Reaction Type(s): <u>Decomposition</u>       |

# Lab 25 Equation Writing And Predicting Products

## Answer Key

**G Thomas**



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at a water depth of 60 m the platform is modified based on the csc braceless concept proposed by luan which is equipped with a 5 mw wind turbine and a catenary mooring system luan designed the fowt for a water depth of 200 m

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