

Name: \_\_\_\_\_

## Double Replacement Lab

Reminder – Goggles must be worn at all times in the lab!

### PRE-LAB DISCUSSION:

A PRECIPITATE is an insoluble substance formed by the reaction between chemicals in solution. A list of reactions will be given to you. The reactants are on the left side of the equation, and the products are on the right. The precipitate will be found on the product side. Keep in mind that an equation gives a condensed picture of what happens when two chemicals react.

### PURPOSE:

The purpose of the lab is to observe semi-micro scale precipitation reactions, and to use a solubility table to identify the insoluble product, when there is one.

### PROCEDURE:

Predict the products and balance the following double replacement reactions. Then, identify the precipitates using the table on page 2. **You may want to do your work step on a scratch piece of paper and attach to lab later.**

<u>Double Replacement Reactions</u>					<b>Precipitates (if any)</b>
A.	NaCl	+	AgNO <sub>3</sub>	→	
B.	KI	+	AgNO <sub>3</sub>	→	
C.	NaOH	+	AgNO <sub>3</sub>	→	
D.	Na <sub>3</sub> PO <sub>4</sub>	+	AgNO <sub>3</sub>	→	
E.	NaCl	+	Pb(NO <sub>3</sub> ) <sub>2</sub>	→	
F.	KI	+	Pb(NO <sub>3</sub> ) <sub>2</sub>	→	
G.	NaOH	+	Pb(NO <sub>3</sub> ) <sub>2</sub>	→	
H.	Na <sub>3</sub> PO <sub>4</sub>	+	Pb(NO <sub>3</sub> ) <sub>2</sub>	→	
I.	NaCl	+	CuSO <sub>4</sub>	→	
J.	KI	+	CuSO <sub>4</sub>	→	
K.	NaOH	+	CuSO <sub>4</sub>	→	
L.	Na <sub>3</sub> PO <sub>4</sub>	+	CuSO <sub>4</sub>	→	

# Lab 10 2 Double Replacement Reactions

**David Baud**



## Lab 10 2 Double Replacement Reactions:

**CHEMISTRY EXPERIMENTS** James Signorelli, 2014-09-19 Gifted and talented students and any student interested in pursuing a science major in college needs a rigorous program to prepare them while they are still in high school This book utilizes a format where the application of several disciplines science math and language arts principles are mandated Each lab concludes with either an essay or a detailed analysis of what happened and why it happened This format is based on the expectations of joining a university program or becoming an industrial science professional The ideal student lab report would be written in a lab research notebook and then the essay or final analysis is done on a word processor to allow for repeat editing and corrections The research notebook has all graph pages a title section and a place for the students and their assistants to sign and witness that exercise The basic mechanics of the lab report title purpose procedure diagrams data table math and calculations observations and graphs are handwritten into the book The conclusion is done on a word processor MS Word which allows the instructor to guide the student in writing and editing a complete essay using the MLA format When the final copy is completed the essay is printed and inserted into the lab notebook for grading At the end of the term the student has all their labs in one place for future reference These lab notebooks can be obtained for as little as 3 00 per book This is money well spent In our district the Board of Education buys the books for each student The BOE sees these books as expendable but necessary materials for all science and engineering instruction **Chemistry** McGraw-Hill Staff, 2001-07 General Chemistry Donald A. McQuarrie, Stanley Gill, 2011-06-15 This Fourth Edition of McQuarrie s classic text offers a thorough revision and a quantum leap forward from the previous edition Taking an atoms first approach it promises to be another ground breaking text in the tradition of McQuarrie s many previous works This outstanding new text available in a soft cover edition offers professors a fresh choice and outstanding value **E-chemistry Iii Tm (science and Technology)' 2003 Ed.** , Laboratory Experiments for General, Organic & Biochemistry Frederick A. Bettelheim, Joseph Marvin Landesberg, 1997 CHEMICAL REACTIONS NARAYAN CHANGDER, 2024-04-08 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in quiz format on our youtube channel <https://www.youtube.com/smartquiziz> I will send you a PDF version of this workbook This book has been designed for candidates preparing for various competitive examinations It contains many objective questions specifically designed for different exams Answer keys are provided at the end of each page It will undoubtedly serve as the best preparation material for aspirants This book is an engaging quiz eBook for all and offers something for everyone This book will satisfy the curiosity of most students while also challenging their trivia skills and introducing them to new information Use this invaluable book to test your subject matter expertise Multiple choice exams are a common assessment method that all prospective candidates must be familiar with in today s academic environment Although the majority of students are accustomed to this MCQ format many are not well versed in it To achieve success in MCQ tests quizzes and

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*Laboratory Experiments for Basic Chemistry* Alan Sherman, Sherman, 1988 Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application

*Safety-Scale Lab Exp Biochem 2e* Spencer L. Seager, Michael R. Slabaugh, 1994-05 **General, Organic, and Biological Chemistry Study Guide and Selected Solutions** Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems

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