#### Cibernical Kinetics

 Molecular hydrogen can be made from methane gas by the reaction below. How is the rate of disappearance of CH<sub>s</sub> related to the rate of appearance of H<sub>a</sub>?

- 20
- 製
- D) none of these

Acres B

25. For the reaction shows below, what is the relationship-between the rate of formation of Br<sub>2</sub>, and the rate of reaction of Br<sup>2</sup> over the same time period?

$$CIO_{2}^{-}(aq) + 4 Br^{-}(aq) + 4 H^{-}(aq) - CI^{-}(aq) + 2 Br_{2}(aq) + 2 H_{2}O()$$

- A) rate of formation of  $\mathrm{Br}_{s_0}=1/2\times\mathrm{rate}$  of reaction of  $\mathrm{Br}$
- B) rate of formation of Br., = 1/4 = rate of reaction of Br.
- C) rate of formation of Er  $_2$  = 2 × rate of reaction of Er  $^{\prime}$
- D) rate of formation of Br  $_{\gamma}$  = 2 × rate of reaction of Br

America.

3). The reaction that occurs in a Breethalytee, a device used to determine the alcohol level in a person's bloodstream, is given below. If the rate of appearance of Gr<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> is 1.24 mol/min at a particular moment, what is the rate of disappearance of G<sub>2</sub>H<sub>2</sub>O at that moment?

- A) 0.413 mol/min
- \$0.0026 molitrán
- C) 1.86 mol/min.
- D) 3.72 mol/min

Asserted C

# **General Equilibrium Ap Chemistry Multiple Choice**

**A Loxley** 

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