

# **GENERAL CHEMISTRY 2**

## **STUDY GUIDE**

authored by: John Michael F. Aguilar

### **What is Kinetic Energy?**

-Kinetic Energy is energy of motion.

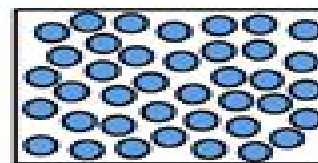
If an object has mass and velocity it has Kinetic Energy. The more Kinetic Energy a particle has the faster it will move, Less Kinetic Energy has the slower it will move.

### **What is Kinetic Molecular Theory?**

-The Kinetic Molecular theory is the theory that all matter is made up of atoms and that are always moving.

These particles hold Kinetic Energy and move around in a random direction. It says that in higher temperature the particles have high energy and in the Low Temperature, the particles have low energy. The more Energy a particle has, the faster it will move.

### **What is the kinetic Molecular Theory of liquid??**



### **How do solid have kinetic energy?**

-The faster the vibration and the particles move around, the higher the kinetic energy. Because solids are tightly packed and vibrate in place, they have the lowest kinetic energy. Because liquids have a larger kinetic energy than solids, the particles slide past one other.

## **FORCES OF ATTRACTION**

**THERE ARE TWO KINDS OF FORCES OF ATTRACTIONS THAT OPERATE IN A MOLECULE:**

# General Chemistry Ii Study Guide

**Y Pai**



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