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Gemini Workshop Manual

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radicalform exponentialform $5 \times 2 \times 2 \times 5$ example $8 \times 5 \times 4$ rewrite as a radical

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web we ll define how they work and use them to rewrite exponential expressions in various ways we previously learned about integer powers first positive and then also negative but what does it mean to raise a number to the 2×5 power

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web radical equations are equations in which variables appear under radical symbols $x^2 \times x^1 \times x$ is a radical equation rational equations are equations in which variables can be found in the denominators of rational expressions $\frac{1}{x} \times \frac{1}{2} \times x$

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web a $x^{2 \times 0 \times i \times 1}$ $q^{2 \times a}$ p^k h^u $r^{a \times 0}$ $l^{s \times a \times o \times j \times f}$ $2^{t \times j \times w}$ $6^{a \times 2 \times r}$ $k^{e \times e}$ r^l x^l $z^{c \times g}$ w^a $4^{a \times k \times l}$ 2^l l^0 $r^{w \times i \times v \times g \times c \times h \times p \times t \times l \times s}$ $o^{h \times r}$ $s^{e \times m \times s \times t \times e \times u \times r \times o \times v \times z \times e \times q \times d \times p}$ 7^o $o^{m \times i \times a \times 2 \times d \times k \times e \times k}$ 7^w $l^{i \times j \times t}$ $u^{h \times f}$ $a^{i \times u \times n \times n \times f \times 4 \times i \times b \times n}$ $y^{i \times 0 \times t \times 2 \times e}$ $u^{g \times a \times h \times l \times g \times g \times b \times e \times 4 \times b \times l \times r}$ $g^{a \times j}$ n^2 y^i worksheet by kuta software llc

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web if you have the same base the product of that base raised to one exponent and that same base raised to another exponent that s the same thing as that base raised to the sum of those exponents a classic exponent property

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exponents radicals are another way to write rational exponents for example $x^{1/2}$ and \sqrt{x} are equivalent in this lesson we'll review the rules of exponent operations with integer exponents

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web consider the thermochemical equation below

$$4\text{NO(g)} + 6\text{H}_2\text{O(l)} + 4\text{NH}_3\text{(g)} + 5\text{O}_2\text{(g)} \rightarrow 11\text{H}_2\text{O(l)} + 4\text{HNO}_3\text{(aq)}$$
 is this reaction endothermic or exothermic
 endothermic
 how much heat will be absorbed or released if 10 g of NO is reacted with 10 g of H₂O

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web heats of fusion and vaporization determines what determines how much energy is required to change a substance's phase
 what is the heat of fusion formula
 $q = m \cdot \Delta H_{\text{fus}}$
 what is the heat of vaporization formula
 $q = m \cdot \Delta H_{\text{vap}}$
 what is on the x-axis of the heating curve
 time

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 3 carry out a thermochemical calculation on all these species at their optimized geometries
 4 sum up the enthalpies free energy of the products and the reagents keeping into account the

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web answer a 1 explanation the entire reaction was multiplied by 2 therefore ΔH must also be halved
 3 answer c explanation
 option c is incorrect because 566 kJ/mol is the ΔH per mole of O₂ and not for 2 moles even though options b and d are expressed in grams they are still correct because 56

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 what is the heat change when 472 g of carbon reacts with excess O₂ according to the following equation

$$\text{C(s)} + \text{O}_2\text{(g)} \rightarrow \text{CO}_2\text{(g)} \quad \Delta H = -393.5 \text{ kJ/mol}$$
 how much heat energy must be expelled by the body to rid the body of one pound of fat
 2830 J
 is this endothermic or exothermic
 exothermic

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 exothermic transfer of heat from the system to the surroundings
 $2\text{H}_2\text{(g)} + \text{O}_2\text{(g)} \rightarrow 2\text{H}_2\text{O(l)}$
 energy endothermic the transfer of heat from the surroundings to the system
 $2\text{Hg(s)} + 2\text{Hg(l)} + \text{O}_2\text{(g)}$

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 enthalpy is an extensive property depends on amount of matter present this fact means that the magnitude of h is directly proportional to the amount of reactant consumed in the process

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web q 140 g 4 184 J 160°C g 0°C q 9372 J 16 J 35.5 g x 1 mole KCN 0.5453 moles 65.1 g 0.5453 mole moles 5
 when a 15 g sample of SrCl₂ dissolves in 80 g of water in a calorimeter the temperature drops from 42°C to 15°C calculate the ΔH for the process

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