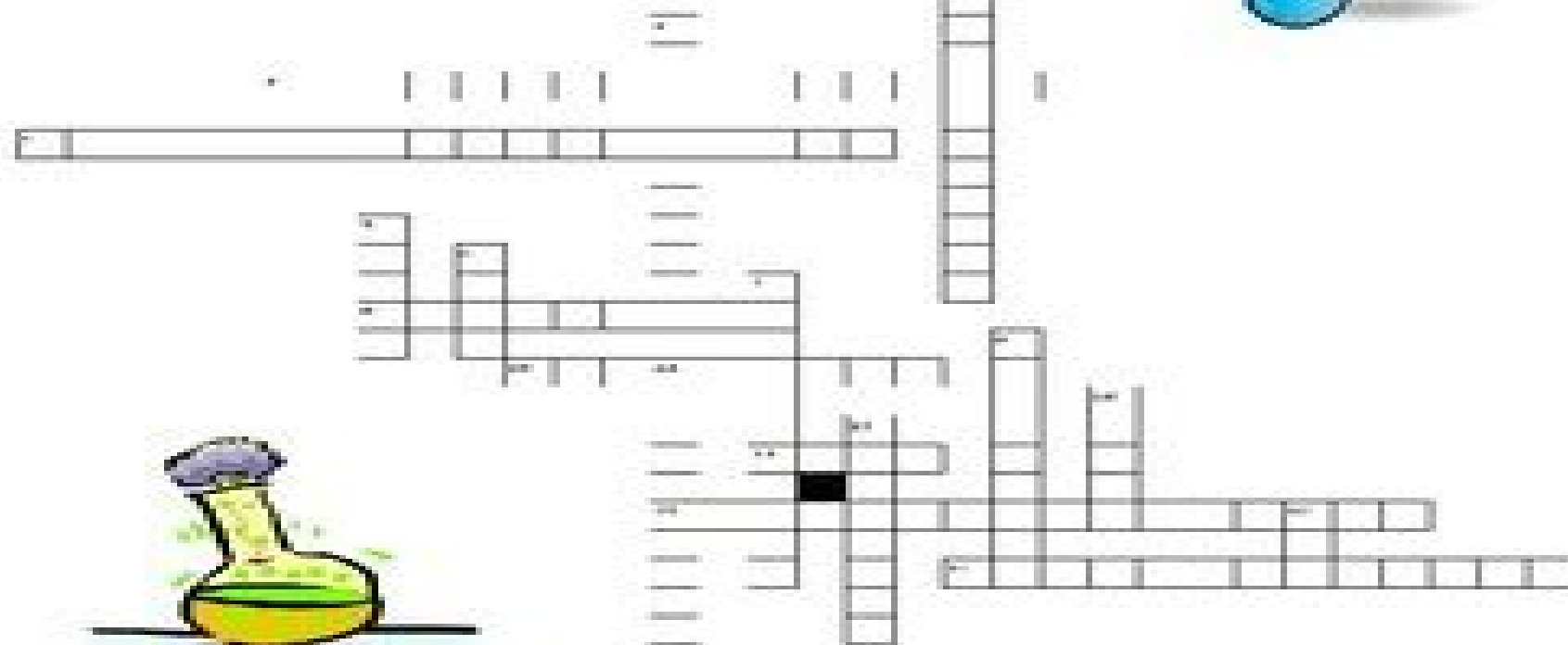




Across

3. A chemical equation that has the same amount of each element on both sides of the equation.
4. The amount of mass that is in the reactions should match the amount of mass in the products.
5. The relationship between the amount of moles in a chemical reaction.
6. The mass of a single element or compound in one mole.
14. A basic unit of a chemical element.
15. A way to changing from one set of units to another.

Down

1. The amount of molecules in one mole.
2. The number in front of a chemical equation, used to balance.
5. How we measure the mass of a molecule.
6. Equal to 6.02×10^{23} .
7. The chemical formula for an ionic compound.
9. The amount of a particular elements or molecules in a chemical equation.
11. The mass of all the protons, neutrons, and electrons in an element.
12. The amount of moles in 126 g of water.
13. A group of two or more atoms.
16. The amount of moles in 32 g of oxygen.

G Stoichiometry Crossword Answers

Evelyn Biluk



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