

Q2) Attempt any four of the following:

[16]

- a) Describe the nature of liquid according to kinetic molecular model.
- b) Explain physical or Vander Waal's adsorption.
- c) Write the equation of the line passing through a point
 - i) $(-4, 0)$ and having slope - 3. &
 - ii) $(-2, 0)$ & slope - $\frac{1}{3}$.
- d) Explain de-Broglies hypothesis and derive the expression for wavelength in terms of kinetic energy of particle.
- e) Explain the importance of Carnot's cycle in terms of
 - i) determination of efficiency.
 - ii) thermodynamic scale of temperature.
- f) Give assumptions and limitations of Bohr's theory.

Q3) Answer any four of the following:

[16]

- a) i) If $Y = \frac{1-x^2}{1+x^2}$, find $\frac{dy}{dx}$.
- ii) Solve the integral $\int (2x)^{\frac{3}{2}} \cdot dx$

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Report on the examination in chemistry, 1993 **Chemistry** Ying-wai Chan, 1981 **Chemistry Syllabus** Pearson Custom Publishing, 1994-05-01 *Practical Chemistry* J. Howard, 1873 **Chemistry syllabus** Caribbean Examination Council, 1987 **Chemistry for Degree Students B.Sc. Semester - IV (As per CBCS)** Madan R.L., 2017 This textbook has been designed to meet the needs of B Sc Fourth Semester students of Chemistry as per the UGC Choice Based Credit System CBCS With its traditional approach to the subject this textbook lucidly explains principles of chemistry Important topics such as transition elements coordination chemistry crystal field theory kinetic theory of gases liquids solids and chemical kinetics are aptly discussed to give an overview of inorganic and physical chemistry Laboratory work has also been included to help students achieve solid conceptual understanding and learn experimental procedures

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