



Name:

Date:

Student Exploration: Calorimetry Lab

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: calorie, calorimeter, joule, specific heat capacity

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. The Latin word *calor* means "heat," and *meter* comes from the Greek word meaning "to measure." What do you think a **calorimeter** does?

measuring the heat of chemical reactions or physical changes as well as heat capacity

2. Where have you heard the word **calorie** before? What do you think a calorie is?

calorie, a unit of energy or heat variously defined. the calorie was originally defined as the amount of heat required at a pressure of 1 standard atmosphere to raise the temperature of 1 gram of water 1 degree celsius.

Gizmo Warm-up

A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up.

Calorimeters can be used to find a substance's **specific heat capacity**. You will use the *Calorimetry Lab* Gizmo to determine the specific heat capacities of various substances.



1. On the **SIMULATION** pane, select **Copper**. Use the slider to set its **Mass** to 200 g. Set the **Water mass** to 200 g. Check that the **Water temp** is set to 30.0 °C and the copper's **Temp** is 90 °C. Select the **GRAPH** tab, and click **Play** (▶).

- A. What was the **Final temperature** of the copper and the water?
- B. How much did the temperature of the copper change?
- C. How much did the temperature of the water change?

34.96

down 55.04

down 165.04

2. Specific heat capacity can be described as a substance's resistance to temperature changes. Which substance has a greater specific heat capacity, copper or water? Explain.

Explore Learning Calorimetry Lab Gizmo Answer Key

Hongru Du

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Explore Learning Calorimetry Lab Gizmo Answer Key:

This book delves into Explore Learning Calorimetry Lab Gizmo Answer Key. Explore Learning Calorimetry Lab Gizmo Answer Key is a vital topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Explore Learning Calorimetry Lab Gizmo Answer Key, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Explore Learning Calorimetry Lab Gizmo Answer Key
 - Chapter 2: Essential Elements of Explore Learning Calorimetry Lab Gizmo Answer Key
 - Chapter 3: Explore Learning Calorimetry Lab Gizmo Answer Key in Everyday Life
 - Chapter 4: Explore Learning Calorimetry Lab Gizmo Answer Key in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, the author will provide an overview of Explore Learning Calorimetry Lab Gizmo Answer Key. The first chapter will explore what Explore Learning Calorimetry Lab Gizmo Answer Key is, why Explore Learning Calorimetry Lab Gizmo Answer Key is vital, and how to effectively learn about Explore Learning Calorimetry Lab Gizmo Answer Key.
 3. In chapter 2, the author will delve into the foundational concepts of Explore Learning Calorimetry Lab Gizmo Answer Key. This chapter will elucidate the essential principles that must be understood to grasp Explore Learning Calorimetry Lab Gizmo Answer Key in its entirety.
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 6. In chapter 5, this book will draw a conclusion about Explore Learning Calorimetry Lab Gizmo Answer Key. This chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Explore Learning Calorimetry Lab Gizmo Answer Key.

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Table of Contents Explore Learning Calorimetry Lab Gizmo Answer Key

1. Understanding the eBook Explore Learning Calorimetry Lab Gizmo Answer Key
 - The Rise of Digital Reading Explore Learning Calorimetry Lab Gizmo Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Explore Learning Calorimetry Lab Gizmo Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Explore Learning Calorimetry Lab Gizmo Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Explore Learning Calorimetry Lab Gizmo Answer Key
 - Personalized Recommendations
 - Explore Learning Calorimetry Lab Gizmo Answer Key User Reviews and Ratings
 - Explore Learning Calorimetry Lab Gizmo Answer Key and Bestseller Lists
5. Accessing Explore Learning Calorimetry Lab Gizmo Answer Key Free and Paid eBooks
 - Explore Learning Calorimetry Lab Gizmo Answer Key Public Domain eBooks
 - Explore Learning Calorimetry Lab Gizmo Answer Key eBook Subscription Services
 - Explore Learning Calorimetry Lab Gizmo Answer Key Budget-Friendly Options
6. Navigating Explore Learning Calorimetry Lab Gizmo Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Explore Learning Calorimetry Lab Gizmo Answer Key Compatibility with Devices
 - Explore Learning Calorimetry Lab Gizmo Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Explore Learning Calorimetry Lab Gizmo Answer Key
 - Highlighting and Note-Taking Explore Learning Calorimetry Lab Gizmo Answer Key
 - Interactive Elements Explore Learning Calorimetry Lab Gizmo Answer Key

8. Staying Engaged with Explore Learning Calorimetry Lab Gizmo Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Explore Learning Calorimetry Lab Gizmo Answer Key
9. Balancing eBooks and Physical Books Explore Learning Calorimetry Lab Gizmo Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Explore Learning Calorimetry Lab Gizmo Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Explore Learning Calorimetry Lab Gizmo Answer Key
 - Setting Reading Goals Explore Learning Calorimetry Lab Gizmo Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Explore Learning Calorimetry Lab Gizmo Answer Key
 - Fact-Checking eBook Content of Explore Learning Calorimetry Lab Gizmo Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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