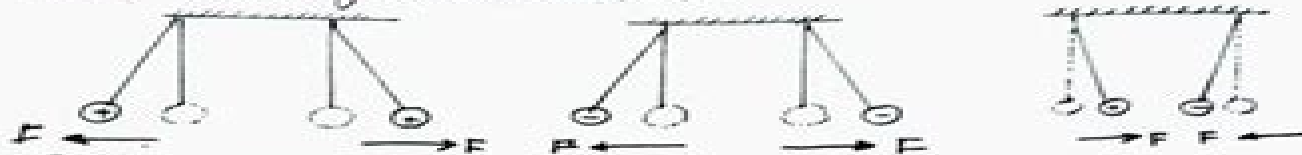


## FIRST LAW OF ELECTROSTATICS

Guericke in 1672 discovered that similar charges have a force of repulsion later after 150 years De Fay found that some time they have the force of attraction. He experimentally proved that

"Similar charges repel each other and opposite charges attract each other" This is called the First law of electrostatics.

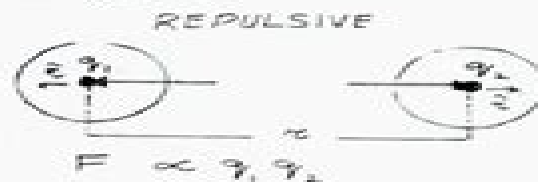
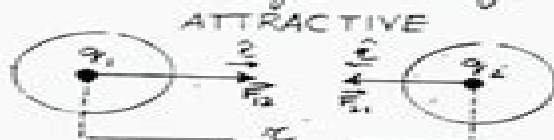


## COULOMBS LAW

Coulomb's law describes the quantitative study of electrostatic forces.

### STATEMENT:

"The force of attraction or repulsion between two point charges is directly proportional to the product of charges and inversely proportional to the square of distance between them."



According to the first part

$$F \propto q_1 q_2$$

According to the second part, called inverse square law

$$F \propto \frac{1}{r^2}$$

Combining the above two

$$F \propto \frac{q_1 q_2}{r^2}$$

$$F = \text{constant} \frac{q_1 q_2}{r^2}$$

$$F = K \frac{q_1 q_2}{r^2}$$

# Electrostatics Physics Test

**Anthony S. Fauci**



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