

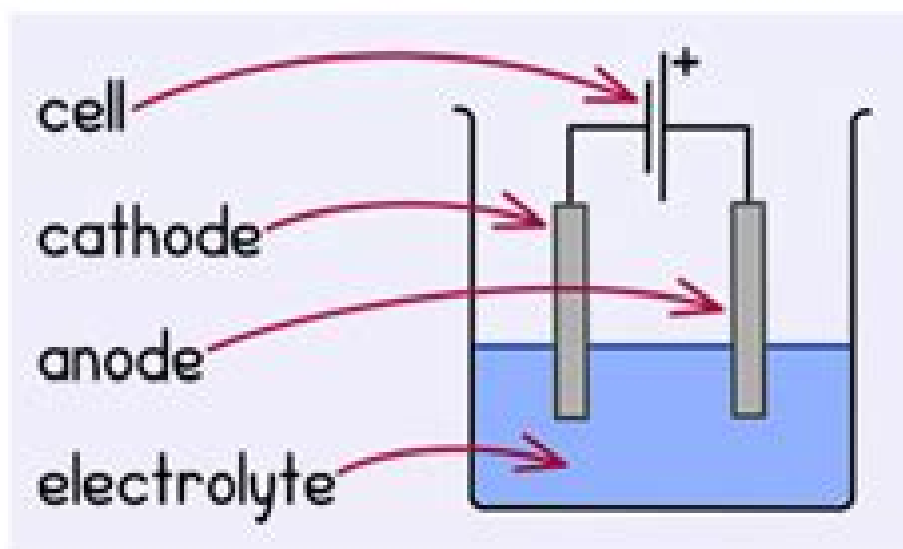
## Electrolysis Required Practical

### Aim

Electrolyse different ionic compounds and identify elements based on your observations.

### Equipment

- Copper (II) chloride solution
- Copper (II) sulphate solution
- Sodium chloride solution
- 100cm<sup>3</sup> beaker
- Two carbon electrodes
- Two leads
- Power supply
- Blue litmus paper
- Tweezers



### Method

1. Measure 50cm<sup>3</sup> of copper (II) chloride solution into a beaker.
2. Plug the leads into the direct current terminals of the power supply and set to 4V.
3. Connect the leads to the two carbon electrodes.

# Electrolysis Study Guide

**Electrolysis Exam Secrets Test Prep**

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