

C2 - THE PERIODIC TABLE 2.6 The transition metals

GCSE Chemistry Aiming for Grade 4

Where are the transition metals on the periodic table?
In the centre block

Give the names of five transition metals

Gold, Silver, Aluminium, Iron, Copper

List the physical properties of the transition metals

1. Good conductors of heat
2. Good conductors of electricity
3. High densities
4. High melting points
5. Hard and strong

Why is mercury not typical as a transition metal?

It is a liquid at room temperature
whereas all the others are solid with
high melting points.



Assessment
Grade 4

GCSE Chemistry Aiming for Grade 6

Compared to Group 1 metals how do the following
properties compare?

Density

Much higher than group 1, none of them will float on
water like lithium, sodium and potassium.

Melting point

Much higher melting points except for mercury

Boiling point

Boiling points are extremely high compared to the
alkali metals

Hardness

Alkali metals can be cut with a knife and are soft.
Transition metals are very hard.

Strength

Transition metals are very strong and can be used for
buildings and machines due to their strength

Give the names of the following transition metal
compounds.

Fe₂O₃

Iron Oxide

Cu₂O

Copper Oxide

NiCO₃

Nickel Carbonate

AgNO₃

Silver Nitrate

Assessment
Grade 6

GCSE Chemistry Aiming for Grade 8

Why is copper used for wires?
It is a very good conductor -
other metals. It is also very
drawn into wires.

Steel is made from Iron and
make girders on bridges?
It is very strong and hard to
weight and high forces.

Titanium is used for aircraft
Titanium has a very high melting
point ideal for very hot engine parts

State the colours of the compounds
of transition metals.

Copper sulphate **blue** due to

Nickel Carbonate **pale green**

Chromium oxide **dark green**

Manganese Chloride **pale pink**

Compared with Group 1 elements
metals have different properties
Transition metals can form a
variety of ions as their reactivity is different.
eg iron can form Fe²⁺ and Fe³⁺
Different reactivity.

Assessment
Grade 8

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1. Understanding the eBook May 2012 Chemistry C2 Mark Scheme
 - The Rise of Digital Reading May 2012 Chemistry C2 Mark Scheme
 - Advantages of eBooks Over Traditional Books
2. Identifying May 2012 Chemistry C2 Mark Scheme
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an May 2012 Chemistry C2 Mark Scheme
 - User-Friendly Interface
4. Exploring eBook Recommendations from May 2012 Chemistry C2 Mark Scheme
 - Personalized Recommendations
 - May 2012 Chemistry C2 Mark Scheme User Reviews and Ratings
 - May 2012 Chemistry C2 Mark Scheme and Bestseller Lists
5. Accessing May 2012 Chemistry C2 Mark Scheme Free and Paid eBooks

- May 2012 Chemistry C2 Mark Scheme Public Domain eBooks
- May 2012 Chemistry C2 Mark Scheme eBook Subscription Services
- May 2012 Chemistry C2 Mark Scheme Budget-Friendly Options
- 6. Navigating May 2012 Chemistry C2 Mark Scheme eBook Formats
 - ePub, PDF, MOBI, and More
 - May 2012 Chemistry C2 Mark Scheme Compatibility with Devices
 - May 2012 Chemistry C2 Mark Scheme Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of May 2012 Chemistry C2 Mark Scheme
 - Highlighting and Note-Taking May 2012 Chemistry C2 Mark Scheme
 - Interactive Elements May 2012 Chemistry C2 Mark Scheme
- 8. Staying Engaged with May 2012 Chemistry C2 Mark Scheme
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers May 2012 Chemistry C2 Mark Scheme
- 9. Balancing eBooks and Physical Books May 2012 Chemistry C2 Mark Scheme
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection May 2012 Chemistry C2 Mark Scheme
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine May 2012 Chemistry C2 Mark Scheme
 - Setting Reading Goals May 2012 Chemistry C2 Mark Scheme
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of May 2012 Chemistry C2 Mark Scheme
 - Fact-Checking eBook Content of May 2012 Chemistry C2 Mark Scheme
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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