

## Type of Chemical Reactions POGIL

**Do Now:** Use this chart below to answer the questions that follow.

I. Synthesis Reaction	II. Decomposition Reaction
 $2\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$	 $\text{H}_2\text{CO}_3 \rightarrow \text{H}_2\text{O} + \text{CO}_2$
III. Single Replacement Reaction	IV. Double Replacement Reaction
 $\text{Zn} + 2\text{HCl} \rightarrow \text{ZnCl}_2 + \text{H}_2$	 $\text{NaCl} + \text{AgI} \rightarrow \text{NaI} + \text{AgCl}$

- What is all the raw materials represented in the chart above called? Are reactants and/or products? Is there any raw materials as products? If yes! Recall that the arrow in a chemical equation is ALWAYS points to the products of the reaction.
- What is all the raw materials represented in the chart above called, and what's happening the products are called? Is it a replacement?

# Lab 14 Types Of Chemical Reactions Answer Key Pearson

**Sebastian Brünink**



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