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# Unit 8 Worksheet 3: Molar Concentration

#### Part A: Describing Molar Concentration

- Sodium chloride was dissolved in water to produce a 1.5M solution. Explain what this concentration tells us about the NaCl solution. How might a chemist use this ratio?
- A 45.3 g sample of potassium nitrate is dissolved in enough water to make 225 mL of solution. Determine the molar concentration of the potassium nitrate.
- Find the molarity of a solution made from 275 g of CuSO<sub>4</sub> dissolved in enough water to make 4.25 L.

 An alcoholic iodine solution ("tincture" of iodine) is prepared by dissolving 5.15 g of iodine crystals in enough alcohol to make a volume of 225 mL. Calculate the molarity of iodine in the solution.

5. What final volume would be needed in order to prepare a 0.25 M NaCl solution from 5.2 g of NaCl (s)?

# **General Chemistry Unit 1worksheet 4 Molar Concentration**

Darla M. Castelli, Leah Fiorentino

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Basics of Analytical Chemistry and Chemical Equilibria Brian M. Tissue, 2013-07-22 Enables students to progressively build and apply new skills and knowledge Designed to be completed in one semester this text enables students to fully grasp and apply the core concepts of analytical chemistry and aqueous chemical equilibria Moreover the text enables readers to master common instrumental methods to perform a broad range of quantitative analyses Author Brian Tissue has written and structured the text so that readers progressively build their knowledge beginning with the most fundamental concepts and then continually applying these concepts as they advance to more sophisticated theories and applications Basics of Analytical Chemistry and Chemical Equilibria is clearly written and easy to follow with plenty of examples to help readers better understand both concepts and applications In addition there are several pedagogical features that enhance the learning experience including Emphasis on correct IUPAC terminology You Try It spreadsheets throughout the text challenging readers to apply their newfound knowledge and skills Online tutorials to build readers skills and assist them in working with the text's spreadsheets Links to analytical methods and instrument suppliers Figures illustrating principles of analytical chemistry and chemical equilibria End of chapter exercises Basics of Analytical Chemistry and Chemical Equilibria is written for undergraduate students who have completed a basic course in general chemistry In addition to chemistry students this text provides an essential foundation in analytical chemistry needed by students and practitioners in biochemistry environmental science chemical engineering materials science nutrition agriculture and the life sciences

Analytical Chemistry Bryan M. Ham, Aihui MaHam, 2015-10-26 A comprehensive study of analytical chemistry providing the basics of analytical chemistry and introductions to the laboratory Covers the basics of a chemistry lab including lab safety glassware and common instrumentation Covers fundamentals of analytical techniques such as wet chemistry instrumental analyses spectroscopy chromatography FTIR NMR XRF XRD HPLC GC MS Capillary Electrophoresis and proteomics Includes ChemTech an interactive program that contains lesson exercises useful calculators and an interactive periodic table Details Laboratory Information Management System a program used to log in samples input data search samples approve samples and print reports and certificates of analysis Quantitative Chemical Analysis Mr. Rohit Manglik, 2024-07-11 EduGorilla Publication is a trusted name in the education sector committed to empowering learners with high quality study materials and resources Specializing in competitive exams and academic support EduGorilla provides comprehensive and well structured content tailored to meet the needs of students across various streams and levels Chemistry Carson-Dellosa Publishing, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill

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categorized by how easy or difficult they are to implement Some for example can work in a large lecture class without a great expenditure of time others may require more preparation and a more adventurous approach to teaching Each strategy is accompanied by a table categorizing its likely impact how much time it will take in class or out and how difficult it will be to implement Like scientific research teaching works best when faculty start with a goal in mind plan an approach building on the literature use well tested methodologies and analyze results for future trials Linda Hodges message is that with such intentional thought and a bit of effort faculty can succeed in helping many more students gain exciting new skills and abilities whether those students are potential scientists or physicians or entrepreneurs. Her book serves as a mini compendium of current research as well as a protocol manual a readily accessible guide to the literature the best practices known to date and a framework for thinking about teaching **Celebrating the Megascale** Phillip J. Mackey, Eric J. Grimsey, Rodney T. Jones, Geoffrey A. Brooks, 2014-01-15 This volume contains the proceedings of the David Robertson Honorary Symposium Celebrating the Megascale held at the TMS 2014Annual Meeting in San Diego California Dr David G C Robertson Professor Emeritus of Metallurgical Engineering at the MissouriUniversity of Science and Technology has devoted his career to the education of highly skilled metallurgical professionals and to the science of all types and sizes of metallurgical processes particularly those involving molten metals The volume contains more than 70 papers covering the important topics and issues in metallurgy today including papers as follows keynote papers covering a tribute to Robertson workforce skillsneeded in the profession going forward copper smelting ladlemetallurgy process metallurgy and resource efficiency new flashiron making technology ferro alloy electric furnace smelting andon the role of bubbles in metallurgical processing operations Topics covered in detail in this volume include ferro alloys non ferrous metallurgy iron and steel modeling education andfundamentals Linne & Ringsrud's Clinical Laboratory Science - E-Book Mary Louise Turgeon, 2015-02-10 Using a discipline by discipline approach Linne perforated pages make it easy for students to remove for use in the lab and for assignment of review questions as homework NEW Instrumentation updates show new technology being used in the lab NEW Additional key terms in each chapter cover need to know terminology NEW Additional tables and figures in each chapter clarify clinical lab science concepts Analytical Chemistry, International Adaptation Gary D. Christian, Purnendu K. Dasgupta, Kevin A. Schug, 2020 With the 7th Edition of Analytical Chemistry renowned chemists Purnendu Sandy Dasgupta and Kevin Schug both of the University of Texas Arlington join the author team The new edition focuses on more in depth coverage of the principles and techniques of quantitative analysis and instrumental analysis aka Analytical Chemistry The goal of the text is to provide a foundation of the analytical process tools and computational methods and resources and to illustrate with problems that bring realism to the practice and importance of analytical chemistry It is designed for undergraduate college students majoring in chemistry and in fields related to chemistry Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science

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