

Quarter 3 Exam Review

Multiple Choice

Always, the choice that best completes the statement or answers the question.

- _____ 1. In a chemical reaction
 - a. the mass of the reactants equals the mass of the products.
 - b. the mass of the product is greater than the mass of reactants.
 - c. the material atoms in the reactants and products must change.
 - d. energy is lost must be added to the reactants.
- _____ 2. When a ratio, provided by a chemical reaction, compares facts the solution it is called
 - a. a percent.
 - b. a constant.
 - c. a percent.
 - d. the mass of the product.
- _____ 3. In writing a chemical equation that produces hydrogen gas, the correct representation of hydrogen gas is
 - a. H
 - b. H₂
 - c. H₂
 - d. H₂
- _____ 4. To balance a chemical equation, it may be necessary to adjust the
 - a. coefficients.
 - b. subscripts.
 - c. formulas of the products.
 - d. masses of products.
- _____ 5. According to the law of conservation of mass, the total mass of the reacting substances is
 - a. always more than the total mass of the products.
 - b. always less than the total mass of the products.
 - c. sometimes more and sometimes less than the total mass of the products.
 - d. always equal to the total mass of the products.
- _____ 6. In the chemical equation $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$
 - a. Mg represents the product magnesium.
 - b. Mg represents the reactant magnesium.
 - c. the atomic yield magnesium.
 - d. O₂ represents the product oxygen gas.
- _____ 7. When the reaction $2\text{Fe} + 3\text{Cl}_2 \rightarrow 2\text{FeCl}_3$ is correctly balanced, what is the coefficient of Fe?
 - a. 4
 - b. 3
 - c. 2
 - d. 1
- _____ 8. Which coefficients correctly balance the formula equation $\text{CaCl}_2 + \text{H}_2\text{O} \rightarrow \text{Ca}(\text{OH})_2$?
 - a. 1, 1, 2
 - b. 1, 2, 1
 - c. 1, 1, 1
 - d. 1, 1, 1
- _____ 9. In which kind of reaction do two or more substances combine to form a new compound?
 - a. decomposition reaction
 - b. single reaction
 - c. double displacement reaction
 - d. synthesis reaction
- _____ 10. The equation $\text{A} + \text{B} \rightarrow \text{C} + \text{D}$ is the general equation for a
 - a. synthesis reaction.
 - b. decomposition reaction.
 - c. combination reaction.
 - d. single displacement reaction.

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