

- Q4) a)** Attempt any three of the following : **[12]**
- i) Describe any one method for the purification of colloids.
 - ii) Explain the set of four quantum numbers.
 - iii) Explain positive and negative catalysis with suitable examples.
 - iv) Explain the general properties of colloids.
 - v) Describe Milikans oil drop method for determination of charge of the electron.
- b)** Attempt any one of the following : **[4]**
- i) What is hydrogen bond? What are the essential conditions to form hydrogen bond.
 - ii) What is tetrahedral hybridisation? Explain formation of CH_4 molecule on the basis of hybridisation.
- Q5) a)** Attempt any two of the following : **[6]**
- i) Mention the various isotopes of hydrogen. Explain physical properties of deuterium.
 - ii) Explain formation of N_2 molecule on the basis of overlap of atomic orbitals.
 - iii) Calculate the molarity of solution which contain 10 gms of K_2CO_3 per 1000ml of water
(Given Molecular Weight of $\text{K}_2\text{CO}_3 = 138$)
- b)** Attempt any two of the following : **[10]**
- i) Give postulates of Pauling's-slater theory.
 - ii) 10 ml solution of NaOH containing 2 gms of alkali per litre is exactly neutralised by 15 ml of solution of H_2SO_4 and 30 ml of HCl solution separately? Calculate the strength of the acid in gms per litre.
 - iii) Give the assumptions of VSEPR theory.

+ + + +

Fybsc Chem Syllabus 2015 Pune University

Raffaela Di Napoli



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Syllabus for Chemistry University of Chicago. Department of Chemistry, 1931

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