

# Lab 3: Exploring the Properties of Gases

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MONDAY, 9/27/2021

Pre-Lab Grade (15 pts):

Post-Lab Grade (15 pts):

Author:

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## Objectives:

The purpose of this experiment would be to investigate the different gas laws involving four gas properties: pressure, volume, temperature, and the number of molecules and their relationships. The experiment will address the question whether changing one gas property would affect the other gas properties based on their relationships.

## Procedure:

### Part 1: Pressure and Volume

- ☐ 1. Plug the LabQuest, connect the gas pressure sensor to channel one of the interface, and turn it on.
- ☐ 2. Use a 20 mL syringe and set up the front edge of the piston at the 15 mL mark, which indicates the amount of air trapped inside the syringe.
- ☐ 3. Securely connect the syringe to the valve of the gas pressure sensor.
- ☐ 4. Put LabQuest at time based mode, wait 60 seconds for the system to equilibrate and then push the play button and record your data as run 1.
- ☐ 5. Keep the syringe securely connected to the sensor and push the piston of the syringe to the number of 13 mL indicating the decrease of the volume of the air inside the syringe by 2 mL. Record the pressure value for this volume as run 2 into the lab notebook.
- ☐ 6. Redo step 5 with three different volumes at 11, 9, and 7 mL and record the pressure values respectively into the lab notebook.
- ☐ 7. Use the volume and pressure data collected and create a graph.

### Part 2: Pressure and Absolute Temperature

- ☐ 1. Insert the temperature probe to channel 2 of the interface. Delete the values from experiment 1.
- ☐ 2. Use 600 mL beaker and 125 mL Erlenmeyer flask and construct pressure and absolute temperature experimental setup. Secure tightly the plastic tubing and the Luer-lock valve to the rubber stopper. Close the Erlenmeyer flask with the rubber stopper tightly. The Luer-lock valve should be put in a closed position.
- ☐ 3. Instill 150 mL of water in the 600 mL beaker and holding the flask, submerge it into the water.
- ☐ 4. Fill the beaker with ice to the level of the flask's neck. Keep the rubber stopper dry. Insert the temperature probe into the beaker. The temperature goal is between 0 and 10 °C. After 60 seconds, when temperature and pressure readings stabilized, record the temperature and pressure values into your notebook.
- ☐ 5. Remove the cold water from the beaker and dry the flask.
- ☐ 6. Instill 300 mL of room temperature water in the 600 mL beaker and holding the flask, submerge it into the water. Keep the rubber stopper dry. Insert the temperature probe into the beaker. The temperature goal is between 20 and 30 °C. After 60 seconds, when temperature and pressure readings stabilized, record the temperature and pressure values into your notebook.
- ☐ 7. Remove the cold water from the beaker and dry the flask.

# Exploring The Properties Of Gases Lab Report

**Michal Rosen-Zvi**



## **Exploring The Properties Of Gases Lab Report:**

## Whispering the Secrets of Language: An Mental Quest through **Exploring The Properties Of Gases Lab Report**

In a digitally-driven earth wherever monitors reign great and instant interaction drowns out the subtleties of language, the profound strategies and mental subtleties concealed within phrases usually get unheard. However, located within the pages of **Exploring The Properties Of Gases Lab Report** a captivating fictional treasure sporting with raw feelings, lies an extraordinary quest waiting to be undertaken. Penned by a talented wordsmith, that marvelous opus encourages visitors on an introspective trip, delicately unraveling the veiled truths and profound influence resonating within the very cloth of every word. Within the emotional depths of this moving review, we can embark upon a sincere exploration of the book is primary themes, dissect their captivating publishing model, and yield to the strong resonance it evokes serious within the recesses of readers hearts.

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