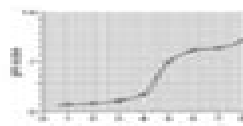


4. State the colour of universal indicator in solutions of:
- an acid _____ (1 mark)
 - an alkali _____ (1 mark)
5. a Do acids taste sweet or sour? _____ (1 mark)
- b Which is dangerous, acid, alkali or they can both be dangerous? _____ (1 mark)

6. Aaf wanted to investigate the pH changes when an acid reacts with an alkali.

He put 5 cm³ of the acid in a tube.
He added the alkali 1 cm³ at a time.
Each time he added the alkali, he measured the pH of the solution.
His results are shown on the graph.



- a. Suggest a way in which Aaf could have measured the pH of the solution. _____ (1 mark)
- b. State the volume of alkali Aaf needed to make the solution neutral. _____ (1 mark)
- c. Explain what the graph shows. _____ (2 marks)

7. A substance has a pH of 12.

State the hazard symbol that should be on a bottle.

_____ (1 mark)

Is the substance an acid or an alkali?

_____ (1 mark)

8. An indicator called bromothymol blue is yellow in an acid and blue in an alkali.
Suggest the colour of bromothymol blue in a neutral solution.

_____ (1 mark)

9. The word equation for a reaction is:



a. State the names of the **two** reactants in the word equation.

_____ (2 marks)

b. State the name of the salt in the word equation.

_____ (1 mark)

10. Hydrochloric acid reacts with calcium oxide to make calcium chloride and water.
Write the word equation for this reaction.

_____ (2 marks)

11. Elen investigated the temperature rise when different acids were added to sodium hydroxide solution.

She measured 10 cm³ of sodium hydroxide solution, and noted the temperature.

She added 10 cm³ of an acid to the sodium hydroxide solution and noted the new temperature.

She calculated the temperature rise.

She repeated the experiment with the same volume of the sodium hydroxide solution but added different acids to it.

Her results are in the table.

Acid added to sodium hydroxide solution	Temperature rise (°C)
nitric acid	5.5
hydrochloric acid	13.0
nitric acid	13.0
sulphuric	8.0

6. Answer the following questions using one of the substances in the box below.

pure water sodium oxide indicator sugar oven cleaner

- a. Which substance contains an acid? _____ (1 mark)
- b. Which substance contains an alkali? _____ (1 mark)
- c. Which substance changes colour to show an acid or an alkali? _____ (1 mark)

7. State what each of these hazard symbols means.

a.  _____ (1 mark)

b.  _____ (1 mark)

c. A bottle of acid is labelled with this hazard symbol.



State two safety precautions you would take before using that acid in an experiment.

Precaution 1: _____ (1 mark)

a. State which variable has been changed in this experiment (the independent variable).

_____ (1 mark)

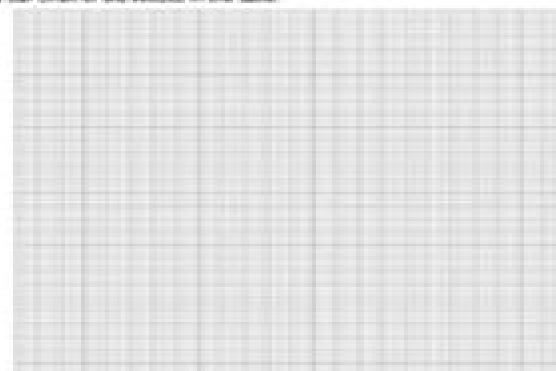
b. State which variable has been measured (the dependent variable).

_____ (1 mark)

c. State the word that describes the reaction between an acid and an alkali.

_____ (1 mark)

d. Put a bar chart of the results in the table.



e. Write a title for the chart.

_____ (1 mark)

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H Kauffman



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1. Understanding the eBook Exploring Science 7 Forces Test Progress Test Mark Scheme
 - The Rise of Digital Reading Exploring Science 7 Forces Test Progress Test Mark Scheme
 - Advantages of eBooks Over Traditional Books
2. Identifying Exploring Science 7 Forces Test Progress Test Mark Scheme
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Exploring Science 7 Forces Test Progress Test Mark Scheme
 - User-Friendly Interface
4. Exploring eBook Recommendations from Exploring Science 7 Forces Test Progress Test Mark Scheme
 - Personalized Recommendations
 - Exploring Science 7 Forces Test Progress Test Mark Scheme User Reviews and Ratings
 - Exploring Science 7 Forces Test Progress Test Mark Scheme and Bestseller Lists
5. Accessing Exploring Science 7 Forces Test Progress Test Mark Scheme Free and Paid eBooks
 - Exploring Science 7 Forces Test Progress Test Mark Scheme Public Domain eBooks
 - Exploring Science 7 Forces Test Progress Test Mark Scheme eBook Subscription Services

- Exploring Science 7 Forces Test Progress Test Mark Scheme Budget-Friendly Options
- 6. Navigating Exploring Science 7 Forces Test Progress Test Mark Scheme eBook Formats
 - ePub, PDF, MOBI, and More
 - Exploring Science 7 Forces Test Progress Test Mark Scheme Compatibility with Devices
 - Exploring Science 7 Forces Test Progress Test Mark Scheme Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Exploring Science 7 Forces Test Progress Test Mark Scheme
 - Highlighting and Note-Taking Exploring Science 7 Forces Test Progress Test Mark Scheme
 - Interactive Elements Exploring Science 7 Forces Test Progress Test Mark Scheme
- 8. Staying Engaged with Exploring Science 7 Forces Test Progress Test Mark Scheme
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Exploring Science 7 Forces Test Progress Test Mark Scheme
- 9. Balancing eBooks and Physical Books Exploring Science 7 Forces Test Progress Test Mark Scheme
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Exploring Science 7 Forces Test Progress Test Mark Scheme
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Exploring Science 7 Forces Test Progress Test Mark Scheme
 - Setting Reading Goals Exploring Science 7 Forces Test Progress Test Mark Scheme
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Exploring Science 7 Forces Test Progress Test Mark Scheme
 - Fact-Checking eBook Content of Exploring Science 7 Forces Test Progress Test Mark Scheme
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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