

Student Exploration: Ionic Bonds

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

Nate can give Clara a marker

2. Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

Maggie can give Fred and Florence one marker each

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The Ionic Bonds Gizmo™ allows you to explore how **ionic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** () to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



1. Each atom consists of a central nucleus and several **shells** that contain electrons. The outermost electrons are called **valence electrons**. (Inner electrons are not shown.)

How many valence electrons does each atom have?

Sodium: 1

Chlorine: 7

2. Click Pause (). Elements can be classified as **metals** and **nonmetals**. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. **Electron affinity** is a measure of how tightly the valence electrons are held.

A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

sodium

Which is a nonmetal?

chlorine

B. Try moving an electron from the metal to the nonmetal. What happens?

The chlorine atom took the electron and added it to its outermost shell

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DJ Losen



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Table of Contents Explore Learning Student Exploration Unit Ionic Bonds Answers

1. Understanding the eBook Explore Learning Student Exploration Unit Ionic Bonds Answers
 - The Rise of Digital Reading Explore Learning Student Exploration Unit Ionic Bonds Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Explore Learning Student Exploration Unit Ionic Bonds Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Explore Learning Student Exploration Unit Ionic Bonds Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Explore Learning Student Exploration Unit Ionic Bonds Answers

- Personalized Recommendations
 - Explore Learning Student Exploration Unit Ionic Bonds Answers User Reviews and Ratings
 - Explore Learning Student Exploration Unit Ionic Bonds Answers and Bestseller Lists
5. Accessing Explore Learning Student Exploration Unit Ionic Bonds Answers Free and Paid eBooks
 - Explore Learning Student Exploration Unit Ionic Bonds Answers Public Domain eBooks
 - Explore Learning Student Exploration Unit Ionic Bonds Answers eBook Subscription Services
 - Explore Learning Student Exploration Unit Ionic Bonds Answers Budget-Friendly Options
 6. Navigating Explore Learning Student Exploration Unit Ionic Bonds Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Explore Learning Student Exploration Unit Ionic Bonds Answers Compatibility with Devices
 - Explore Learning Student Exploration Unit Ionic Bonds Answers Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Explore Learning Student Exploration Unit Ionic Bonds Answers
 - Highlighting and Note-Taking Explore Learning Student Exploration Unit Ionic Bonds Answers
 - Interactive Elements Explore Learning Student Exploration Unit Ionic Bonds Answers
 8. Staying Engaged with Explore Learning Student Exploration Unit Ionic Bonds Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Explore Learning Student Exploration Unit Ionic Bonds Answers
 9. Balancing eBooks and Physical Books Explore Learning Student Exploration Unit Ionic Bonds Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Explore Learning Student Exploration Unit Ionic Bonds Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Explore Learning Student Exploration Unit Ionic Bonds Answers
 - Setting Reading Goals Explore Learning Student Exploration Unit Ionic Bonds Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Explore Learning Student Exploration Unit Ionic Bonds Answers

- Fact-Checking eBook Content of Explore Learning Student Exploration Unit Ionic Bonds Answers
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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