

Name: _____ Date: _____

Student Exploration: Nuclear Decay

Vocabulary: alpha particle, atomic number, beta particle, daughter product, gamma ray, isotope, mass number, nuclear decay, positron, radioactive, subatomic particle

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

The chart below gives the locations, charges, and approximate masses of three **subatomic particles**. The approximate mass of each particle is given in universal mass units (u).

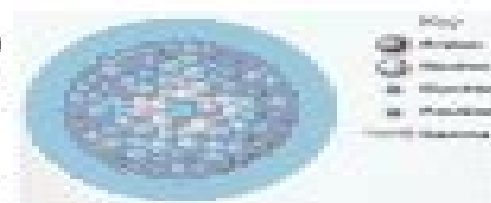
Particle	Location	Charge	Approximate mass
Proton	Nucleus	1^+	1 u
Neutron	Nucleus	0	1 u
Electron	Orbitals	1^-	0 u

- The **mass number** of an atom is equal to the sum of protons and neutrons in the nucleus. A helium atom has 2 protons and 2 neutrons. What is the mass number of this atom? _____
- The **atomic number** of an element is equal to the number of protons in each atom of the element. All helium atoms have 2 protons. What is the atomic number of helium? _____

Gizmo Warm-up

While most atoms are stable, some are **radioactive**, which means that they have a tendency to undergo spontaneous **nuclear decay**. The decay of radioactive atoms generally results in the emission of particles and/or energy.

Several types of nuclear decay can be explored with the Nuclear Decay Gizmo™. On the Gizmo, check that **Alpha decay** and **Uranium** are selected.



- Click **Play** (▶) and then click **Pause** (⏸) when the **alpha particle** is clearly visible. What is an alpha particle made of? _____
- Click **Play** and observe. Besides the alpha particle, what else is emitted from the nucleus during alpha decay? _____

Gamma rays are energetic electromagnetic waves; they are often emitted in nuclear decay.

Explore Learning Student Exploration Nuclear Decay Answers

JR Anderson



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Table of Contents Explore Learning Student Exploration Nuclear Decay Answers

1. Understanding the eBook Explore Learning Student Exploration Nuclear Decay Answers
 - The Rise of Digital Reading Explore Learning Student Exploration Nuclear Decay Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Explore Learning Student Exploration Nuclear Decay Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Explore Learning Student Exploration Nuclear Decay Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Explore Learning Student Exploration Nuclear Decay Answers
 - Personalized Recommendations
 - Explore Learning Student Exploration Nuclear Decay Answers User Reviews and Ratings

- Explore Learning Student Exploration Nuclear Decay Answers and Bestseller Lists
- 5. Accessing Explore Learning Student Exploration Nuclear Decay Answers Free and Paid eBooks
 - Explore Learning Student Exploration Nuclear Decay Answers Public Domain eBooks
 - Explore Learning Student Exploration Nuclear Decay Answers eBook Subscription Services
 - Explore Learning Student Exploration Nuclear Decay Answers Budget-Friendly Options
- 6. Navigating Explore Learning Student Exploration Nuclear Decay Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Explore Learning Student Exploration Nuclear Decay Answers Compatibility with Devices
 - Explore Learning Student Exploration Nuclear Decay Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Explore Learning Student Exploration Nuclear Decay Answers
 - Highlighting and Note-Taking Explore Learning Student Exploration Nuclear Decay Answers
 - Interactive Elements Explore Learning Student Exploration Nuclear Decay Answers
- 8. Staying Engaged with Explore Learning Student Exploration Nuclear Decay Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Explore Learning Student Exploration Nuclear Decay Answers
- 9. Balancing eBooks and Physical Books Explore Learning Student Exploration Nuclear Decay Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Explore Learning Student Exploration Nuclear Decay Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Explore Learning Student Exploration Nuclear Decay Answers
 - Setting Reading Goals Explore Learning Student Exploration Nuclear Decay Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Explore Learning Student Exploration Nuclear Decay Answers
 - Fact-Checking eBook Content of Explore Learning Student Exploration Nuclear Decay Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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