

Names & Formulas for Ionic Compounds

Give the name or formula of the following ionic compounds:

Name		Formula	
1)	<chem>Na2CO3</chem> _____	21)	sodium phosphide _____
2)	<chem>NaOH</chem> _____	22)	magnesium nitrate _____
3)	<chem>MgBr2</chem> _____	23)	lead (II) sulfide _____
4)	<chem>KCl</chem> _____	24)	calcium phosphate _____
5)	<chem>PbCl2</chem> _____	25)	ammonium sulfate _____
6)	<chem>PbCl2</chem> _____	26)	silver cyanide _____
7)	<chem>Zn(OH)2</chem> _____	27)	aluminum sulfide _____
8)	<chem>Na2SO4</chem> _____	28)	barium chloride _____
9)	<chem>CuF2</chem> _____	29)	copper (II) arsenate _____
10)	<chem>Al2S3</chem> _____	30)	iron (III) oxide _____
11)	<chem>PbO</chem> _____	31)	gallium nitride _____
12)	<chem>Li3PO4</chem> _____	32)	iron (II) carbonate _____
13)	<chem>TiCl4</chem> _____	33)	samarium (IV) phosphide _____
14)	<chem>Ca3P2</chem> _____	34)	calcium oxide _____
15)	<chem>Mg3P2</chem> _____	35)	magnesium acetate _____
16)	<chem>Ca(NO3)2</chem> _____	36)	aluminum sulfate _____
17)	<chem>Ag2SO4</chem> _____	37)	copper (II) carbonate _____
18)	<chem>Na2CO3</chem> _____	38)	barium oxide _____
19)	<chem>AgCN</chem> _____	39)	ammonium sulfide _____
20)	<chem>Na2C2O4</chem> _____	40)	silver bromide _____

Naming Balancing And Formula Writing Worksheet

**Barron's Educational Series, Neil D.
Jespersen, Pamela Kerrigan**



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