

# Chemistry Practice Writing Chemical Formulas

Write a chemical formula for each compound.

1.	<u><math>\text{Al}_2\text{O}_3</math></u>	aluminum oxide
2.	<u><math>\text{H}_2\text{SO}_4</math></u>	sulfuric acid
3.	<u><math>\text{H}_2\text{S}</math></u>	hydrogen sulfide
4.	<u><math>\text{K}_2\text{SO}_4</math></u>	potassium sulfate
5.	<u><math>\text{H}_2\text{CO}_3</math></u>	carbonic acid
6.	<u><math>\text{H}_2\text{SO}_3</math></u>	sulfurous acid
7.	<u><math>\text{Ca}_3(\text{PO}_4)_2</math></u>	calcium phosphate
8.	<u><math>\text{Mg}_3\text{N}_2</math></u>	magnesium nitride
9.	<u><math>\text{Ca}_3(\text{PO}_4)_2</math></u>	calcium phosphate
10.	<u><math>\text{H}_2\text{SO}_4</math></u>	sulfuric acid
11.	<u><math>\text{H}_2\text{SO}_3</math></u>	sulfurous acid
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80.	<u><math>\text{H}_2\text{SO}_4</math></u>	sulfuric acid

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