

Analytical Chemistry MCQs: TYBSC SEM III

1] Gravimetric Analysis

1. If the solubility of calcium fluoride in water is S moles per litre, its solubility is given by

- (a) $2S$ (b) $2S^2$
(c) $4S^3$ (d) $4S^2$

2. Correct representation of solubility product of $\text{Ca}_3(\text{PO}_4)_2$ is

- (a) $[\text{Ca}^{2+}][\text{PO}_4^{3-}]^2$ (b) $[\text{Ca}^{2+}][\text{PO}_4^{3-}]$
(c) $[\text{Ca}][\text{PO}_4]$ (d) none of these

3. Dimethyl glyoxime is a reagent specific for the metal ion.

- (a) Al (b) Ni
(c) Fe (d) Cu

4. The solubility product expression for Hg_2Cl_2 is

- (a) $[\text{Hg}^{2+}][\text{Cl}^-]$ (b) $[\text{Hg}_2^{2+}][\text{Cl}^-]^2$
(c) $[\text{Hg}_2^{2+}][2\text{Cl}^-]$ (d) $[\text{Hg}^{+1}][\text{Cl}^-]$

5. A solution is called saturated if

- (a) ionic product < solubility product.
(b) ionic product > solubility product.
(c) ionic product = solubility product.
(d) none of these

6. Solubility of alcohol in water is due to

- (a) ionic bonding
(b) covalent bonding
(c) Metallic bonding
(d) Hydrogen bonding

7. Precipitation takes place when the product of concentration of ions

- (a) Exceeds their solubility product
(b) Is less than their solubility product
(c) Equal to their solubility product
(d) None of these

Mcq Of Analytical Chemistry

RJ Shavelson



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Table of Contents Mcq Of Analytical Chemistry

1. Understanding the eBook Mcq Of Analytical Chemistry
 - The Rise of Digital Reading Mcq Of Analytical Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Mcq Of Analytical Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mcq Of Analytical Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mcq Of Analytical Chemistry
 - Personalized Recommendations
 - Mcq Of Analytical Chemistry User Reviews and Ratings
 - Mcq Of Analytical Chemistry and Bestseller Lists
5. Accessing Mcq Of Analytical Chemistry Free and Paid eBooks
 - Mcq Of Analytical Chemistry Public Domain eBooks
 - Mcq Of Analytical Chemistry eBook Subscription Services
 - Mcq Of Analytical Chemistry Budget-Friendly Options

6. Navigating Mcq Of Analytical Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Mcq Of Analytical Chemistry Compatibility with Devices
 - Mcq Of Analytical Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mcq Of Analytical Chemistry
 - Highlighting and Note-Taking Mcq Of Analytical Chemistry
 - Interactive Elements Mcq Of Analytical Chemistry
8. Staying Engaged with Mcq Of Analytical Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mcq Of Analytical Chemistry
9. Balancing eBooks and Physical Books Mcq Of Analytical Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mcq Of Analytical Chemistry
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Mcq Of Analytical Chemistry
 - Setting Reading Goals Mcq Of Analytical Chemistry
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mcq Of Analytical Chemistry
 - Fact-Checking eBook Content of Mcq Of Analytical Chemistry
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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