

GRADE 6 BENCHMARK ASSESSMENTS ANSWER KEYS

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Test 1

Question	Answer	Content Focus	CCSS	Complexity
1A	C	Characters, Setting, Plot: Sequence	RL.6.3	DOK 3
1B	see below	Characters, Setting, Plot: Sequence/ Text Evidence	RL.6.3/RL.6.1	DOK 3
2A	A	Personification	L.6.5a	DOK 2
2B	D	Personification/Text Evidence	L.6.5a/RL.6.1	DOK 2
3	see below	Characters, Setting, Plot: Compare and Contrast	RL.6.3	DOK 2
4	see below	Theme	RL.6.2	DOK 3
5A	B	Author's Point of View	RI.6.6	DOK 3
5B	A	Author's Point of View/Text Evidence	RI.6.6/RI.6.1	DOK 3
6A	B	Main Idea and Key Details	RI.6.2	DOK 2
6B	D, F	Main Idea and Key Details/ Text Evidence	RI.6.2/RI.6.1	DOK 2
7A	C	Text Features: Maps	RI.6.7	DOK 2
7B	B	Text Features: Maps/Text Evidence	RI.6.7/RI.6.1	DOK 2
8A	D	Context Clues: Paragraph Clues	L.6.4a	DOK 2
8B	C	Context Clues: Paragraph Clues/ Text Evidence	L.6.4a/RI.6.1	DOK 2
9	see below	Text Structure: Problem and Solution	RI.6.2	DOK 2
10	A	Greek and Latin Prefixes	L.6.4b	DOK 1
11	A, B, E	Research	W.6.1b	DOK 3
12	see below	Research	W.6.8	DOK 2
13	A	Research	W.6.8	DOK 2
14	D	Research	W.6.8	DOK 2
15A	A	Theme	RL.6.2	DOK 3
15B	D, F	Theme/Text Evidence	RL.6.2/RL.6.1	DOK 3
16A	B	Metaphors and Similes	L.6.5a	DOK 2
16B	D	Metaphors and Similes/Text Evidence	L.6.5a/RL.6.1	DOK 2
17A	C	Connotation and Denotation	L.6.5c	DOK 2
17B	B	Connotation and Denotation/ Text Evidence	L.6.5c/RL.6.1	DOK 2
18A	D	Characters, Setting, Plot: Sequence	RL.6.3	DOK 3
18B	D	Characters, Setting, Plot: Sequence/ Text Evidence	RL.6.3/RL.6.1	DOK 3
19	see below	Point of View	RL.6.6	DOK 3
20	see below	Characters, Setting, Plot: Compare and Contrast	RL.6.3	DOK 3

Mathematics Benchmark Assessment Spring 2014

Answer Key

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subsequent Δh values $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g}) \quad \Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \quad \Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g}) \quad \Delta h = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 2 \text{H}_2\text{O}(\text{l}) \quad \Delta h = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + 7/2 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{l}) \quad \Delta h = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + 1/2 \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta h = 285 \text{ kJ/mole}$

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