SECTION "B" (Short Answer Question) Marks: 24 NOTE: Answer any EIGHT questions from this section.

MINOS

Each question carries 3 marks.

2. Define the following terms:

(i) Periodicity (ii) Electronegativity (iii) Electrolysi

3. Write down three differences between Alkali metals and Alkaline earth metals.

4. The following statements belong to which branch of Chemistry? Identify: (i) That chlorofluoro compounds are responsible for depletion of Ozone laver.

(ii) That investigates the properties of Ammonia gas.

(iii) That find out the quantity and quality of Paracetamol in Panadol tablet.

5. How do you preparer 100ml of 0.4M MgSO4 from a stock solution of 2. OM MgSO4?

6. What is atomic radius? Discuss its trend across the period and group in the periodic table.

7. What are isotope? Describe briefly three isotopes of Uranium.

8. Balance the following chemical equations by inspection methode

(i) CaCO₃ + HCl → CaCl₂ + H₂O + CO₂

(ii) KClO₃ → KCl + O₂ 4/02 - NO + H20

9. Calculate the number of atoms presents in 9.2 g of Calcium. (Atomic mass of Ca = 40)

10. Draw electronic dot and cross structures or following molecules:

(ii) Na (i) H2 (iiii) O2

(Atomic number of H = 1, N = 7, O = 8)

11. Write down the chemical formulae of the following compounds: (i) Caustic Soda

(ii) Sand (iii) Sugar (iv) Lime Stone (v) Gypsum (vi) Bakin; Soda

12. What is meant by Tyndall effect? How is it be used for the identification of a true solution and a colloidal?

13. Justify the following:

(i) Why Fluoride ion is not considered as Neon atom?

(ii) Why noble gases do not react with other elements at ordinary conditions.

Section C (Detailed -Answer Question) 24

NOTE: Attempt any FOUR questions from this section.

Each question carries 6 marks.

14. State Charles law A 600 ml sample of gas is heated from 27 C to 77°C at constant pressure. What would be its final volume?

15. Define electroplating. Describe the Tin plating of an Iron spoon with the help of ionic equations. Draw its labelled diagram also.

16. Identify the highly electronegative group of Periodic table. Enlist the elements present in this group. Give three common characteristics of these elements.

17. Write down six postulates of Bohr's Atomic Model.

18. What is an ionic bond? Explain the formation of ionic bond between Sodium and Chlorine ionic compound Sodium chloride is formed.

19. What are molecules? Discuss its five types with examples.

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Padhraic Smyth

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Table of Contents Karachi Board Chemistry For Class Xii

- 1. Understanding the eBook Karachi Board Chemistry For Class Xii
 - o The Rise of Digital Reading Karachi Board Chemistry For Class Xii
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Karachi Board Chemistry For Class Xii
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Karachi Board Chemistry For Class Xii
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Karachi Board Chemistry For Class Xii
 - Personalized Recommendations

- Karachi Board Chemistry For Class Xii User Reviews and Ratings
- Karachi Board Chemistry For Class Xii and Bestseller Lists
- 5. Accessing Karachi Board Chemistry For Class Xii Free and Paid eBooks
 - Karachi Board Chemistry For Class Xii Public Domain eBooks
 - Karachi Board Chemistry For Class Xii eBook Subscription Services
 - Karachi Board Chemistry For Class Xii Budget-Friendly Options
- 6. Navigating Karachi Board Chemistry For Class Xii eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Karachi Board Chemistry For Class Xii Compatibility with Devices
 - Karachi Board Chemistry For Class Xii Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Karachi Board Chemistry For Class Xii
 - o Highlighting and Note-Taking Karachi Board Chemistry For Class Xii
 - o Interactive Elements Karachi Board Chemistry For Class Xii
- 8. Staying Engaged with Karachi Board Chemistry For Class Xii
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Karachi Board Chemistry For Class Xii
- 9. Balancing eBooks and Physical Books Karachi Board Chemistry For Class Xii
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Karachi Board Chemistry For Class Xii
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Karachi Board Chemistry For Class Xii
 - Setting Reading Goals Karachi Board Chemistry For Class Xii
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Karachi Board Chemistry For Class Xii
 - Fact-Checking eBook Content of Karachi Board Chemistry For Class Xii

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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