

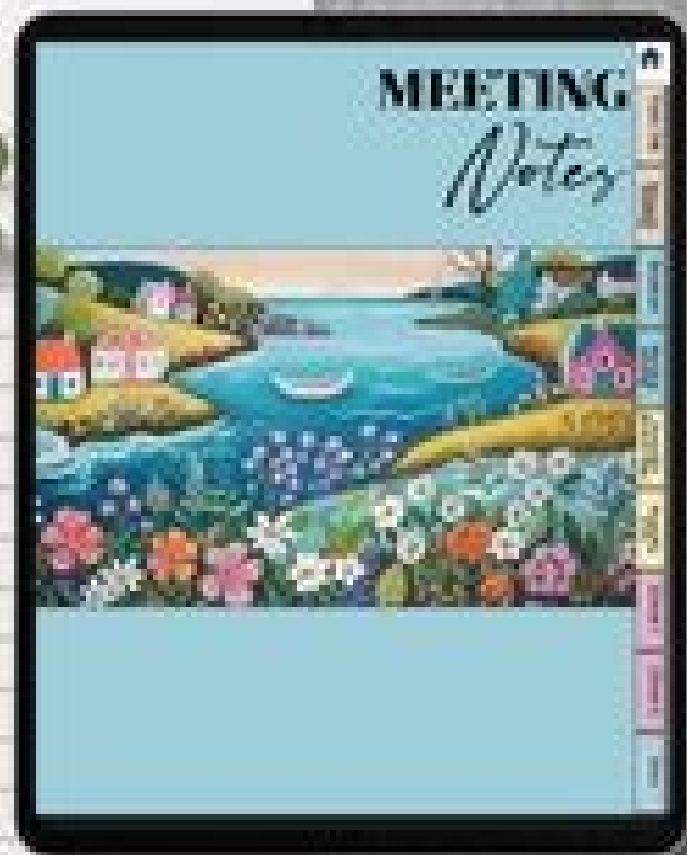
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web one of the ions produced is always H^+ which immediately combines with a water molecule to form the hydronium ion H_3O^+ the H^+ ion is what defines the acidic properties of a substance because of their special classification acids have a naming system different from ionic or other molecular covalent compounds

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particledescriptiontable 1 particle descriptions strong and strong names of various acids particles not drawn to scale Cl^- $ClBr^-$
brf H^+ $HHHHH^+$ HCl HBr $HHFCl$ HBr $HFFormulaHCl$ HBr $HfNameHydrochloric$ acid

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web the molecular or empirical formula acids have their own set of naming rules cf sec 2 8 which you must also know you also need to be able to recognize and name a few organic compounds such as simple alkanes and alcohols see sec 2 9 key questions 8 in the chemical formula and name which element is given first a metal or nonmetal 9

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