NOMENCLATURE PACKET

Worksheet I: Binary Ionic Compounds (representative metals) - metals from groups 1A, 2A, and 3A (1, 2, and 13) have constant charges as ions and do NOT get Roman Numerals in their names

1. Name the following ionic compounds:

a.	Al ₂ O ₃	Aluminum oxide
ь.	Cs2O	Cesium oxide
c.	Rb ₃ N	Rubidium nitride
d.	Ca ₃ N ₂	Calcium nitride
c.	SrSe	Strontium selenide
f.	Cs₂S	Cesium sulfide
9.	Al ₂ S ₃	Aluminum sulfide
h.	LiBr	Lithium bromide
ı.	Mg ₃ N ₂	Magnesium nitride
i	CaF ₂	Calcium fluoride

2. Write the chemical formula for the following ionic compounds:

a.	Barium nitride	Ba*2	N-3	Ba ₃ N ₂
ь.	Indium Fluoride	In-3	F-1	InF ₃
c .	Calcium oxide	Ca-2	0-2	CaO
d.	Sodium nitride	Na ⁻¹	N-3	Na ₃ N
e.	Magnesium chloride	Mg*2	CI-1	MgCl ₂
f.	Potassium oxide	K*1	0.2	K ₂ O
9.	Magnesium oxide	Mg*2	0-2	MgO
h.	Potassium sulfide	K*1	5-2	K25
i	Lithium nitride	Li	N-3	Li ₃ N
J-	Strontium fluoride	5r*2	F-1	SrF2
k.	Aluminum sulfide	A1-3	5-2	Al ₂ S ₃
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Ions And Their Chemical Formula Answers

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