

Properties of Ionic Compounds

Context: Molecules with ionic bonds have many similar properties that occur because of the properties of ionic bonds as well as the crystalline structure of the ionic solids. Because of these common properties of ionic compounds it is important to identify why these properties happen by examining the molecules at the molecular level.

Model 1: Ionic Compound Demonstration Data Table

	NaCl Table Salt	NaHCO ₃ Baking Soda	MgSO ₄ Epsom Salt	CaCO ₃ Limestone /Antacid
Brittle?				
Conducts Electricity as a solid?				
Water Soluble?				
Conducts Electricity in water?				
Largest Electronegativity Difference				

Questions:

1) Identify which two atoms in each molecule would have the largest electronegativity difference. Record this on the data table.

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