HYDROCARBONS



Hydrocarbon Compounds

Hydrocarbon compounds are compounds that consist only of carbon and hydrogen atoms. Hydrocarbons can be divided into 2, namely:

a. Saturated Hydrocarbons: single bonds

including saturated hydrocarbons:

- Alkane: open chain (C_nH_{2n+2})
- a Cycloalkane : closed chain (C_nH_{≥n})

b. Unsaturated Hydrocarbons: double bonds

- Alkene : double bond (C_nH_{2n})
- Alkyne : triple bond (C_nH_{2n-2})
- Alkadiene : two double bonds (C_nH_{2n-2})



Nomenclature of Alkanes

The nomenclature of alkanes is the basis for the nomenclature of other carbon compounds. Therefore the nomenclature of alkanes must be thoroughly understood.

Here are the names of askanes from C1-C10:

Formula	Name	Formula	Name
CH4	methane	C6H14	hexane
C ₂ H ₆	ethane	C7H16	heptane
C ₃ H ₈	propane	C ₈ H ₁₈	octane
C4H10	butane	C9H20	nonane
C ₅ H ₁₂	pentane	C ₁₀ H ₂₂	decane

Atkene names :

 C_2H_4 = ethylene

 C_3H_6 = propene

Hydrocarbons Study Guide

R Bogdan

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Adopting the Melody of Term: An Psychological Symphony within Hydrocarbons Study Guide

In a world eaten by displays and the ceaseless chatter of quick transmission, the melodic beauty and emotional symphony developed by the written term usually disappear into the background, eclipsed by the constant noise and interruptions that permeate our lives. Nevertheless, situated within the pages of **Hydrocarbons Study Guide** a wonderful fictional treasure filled with fresh thoughts, lies an immersive symphony waiting to be embraced. Crafted by a masterful musician of language, that fascinating masterpiece conducts visitors on a psychological journey, skillfully unraveling the concealed melodies and profound affect resonating within each cautiously constructed phrase. Within the depths with this moving analysis, we will discover the book is main harmonies, analyze their enthralling publishing model, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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