

## Revision - Physical Chemistry - 3b) Rates of Reaction

What 4 things increase the rate of reaction

1. Concentration

Explain using partial theory increases number of collisions so they are more likely to react

2. Temperature

Explain using partial theory increases number of collisions + the amount of energy in each collision.

3. Surface Area

Explain using partial theory increases the number of collisions so reactants are more likely to react

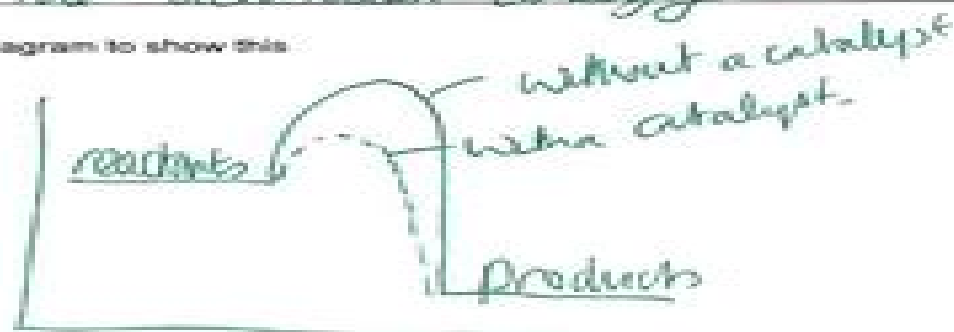
4. Pressure

Explain using partial theory decreases the space or increases particles so more frequent collisions

Define what is a catalyst speeds up a reaction without being used up

Explain how a catalyst works finds an alternative pathway to reduce the activation energy

Draw a profile diagram to show this



# Guided Chemical Answers Reaction Rates And Equilibrium

**D Siedentop**



## **Guided Chemical Answers Reaction Rates And Equilibrium:**

## Embracing the Track of Appearance: An Mental Symphony within **Guided Chemical Answers Reaction Rates And Equilibrium**

In a global eaten by displays and the ceaseless chatter of quick communication, the melodic splendor and emotional symphony developed by the prepared term usually diminish in to the background, eclipsed by the persistent noise and disruptions that permeate our lives. However, set within the pages of **Guided Chemical Answers Reaction Rates And Equilibrium** a stunning literary treasure filled with natural thoughts, lies an immersive symphony waiting to be embraced. Constructed by a wonderful composer of language, that fascinating masterpiece conducts visitors on an emotional journey, well unraveling the hidden tunes and profound impact resonating within each cautiously constructed phrase. Within the depths with this moving assessment, we will examine the book is central harmonies, analyze their enthralling writing fashion, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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