Black to the Troposphere: Acid Rain and Opene

All all districts

Sulfur dioxide is the primary cause of acid rain, although nitrogen oxides and carbon dioxide in the atmosphere can be converted to nitric sold and carbonic sold, respectively. The reactions that occur in the atmosphere to convert sulfur dioxide to sulfuric sold are complicated and involve a hydroxyl free radical. This free radical is a hydroxyle ion missing one electron.

The reaction by which acid rain forms from SO, can occur in a series of three steps:

$$SO_0 + OH \rightarrow HSO_0$$

 $HSO_1 + D_2 \rightarrow SO_1 + HOO$
 $SO_0 + H_2O \rightarrow H_2SO_2$
Oversition: $SO_2 + OH + HSO_2 + O_1 + SO_2 + H_2O \rightarrow HSO_2 + SO_3 + HOO + H_2SO_4(g)$
Cancelling out not intermediates: $SO_1 + OH + O_1 + H_2O \rightarrow HOO + H_2SO_4(g)$

The take home message: suffur dicode reacts with coygen and water in the atmosphere to make suffuric acid, with the help of hydroxyl free radicals.

Ottower.

What is "good" vs. "bad" exerc?

Ocone is produced both in the stratosphere and at ground level. Ozone can be good or bad, depending on where it is found.

Stratospheric opone (good opone) occurs naturally in the upper atmosphere, where it forms a protective layer that sherids us from the sums harmful ultraviolet rays. This beneficial opone has

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A Gutmann

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